

# 晶云药物第二届晶型专题技术培训

## 药物晶型研究和固态研发中的无定形

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# 提纲

- 为什么要研究无定形?
- 无定形基础知识
- 无定形固态分散物制剂
- 无定形固态分散物制剂的制备
- 无定形的检测和表征
- 无定形的稳定性问题



# 为什么要研究无定形？

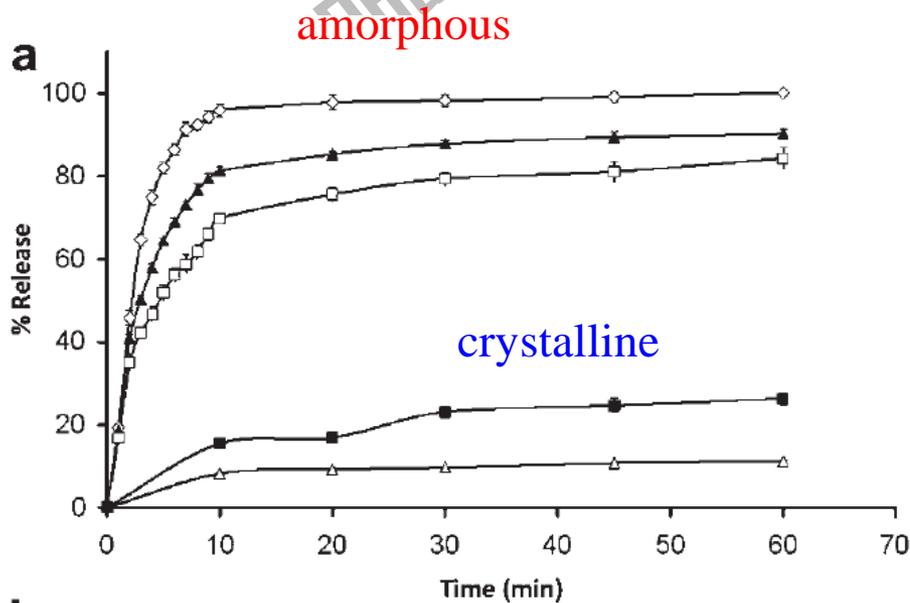


# 制药行业面临的挑战

- Revenues are under pressure
  - Change in target focus
  - Patents expiring
  - Increased regulatory standards
- Drug discovery shift starting in the 1990s:
  - In vitro biology screening approach change
  - Popularity of combinatorial chemistry increased
  - **New Chemical Entities (NCEs) are getting less soluble**
  - **Result: solubility and/or dissolution limited absorption in vivo**
- Significant candidate termination in clinical studies
  - PK/Bioavailability
  - Formulation
  - **Compounds are not achieving adequate exposure**

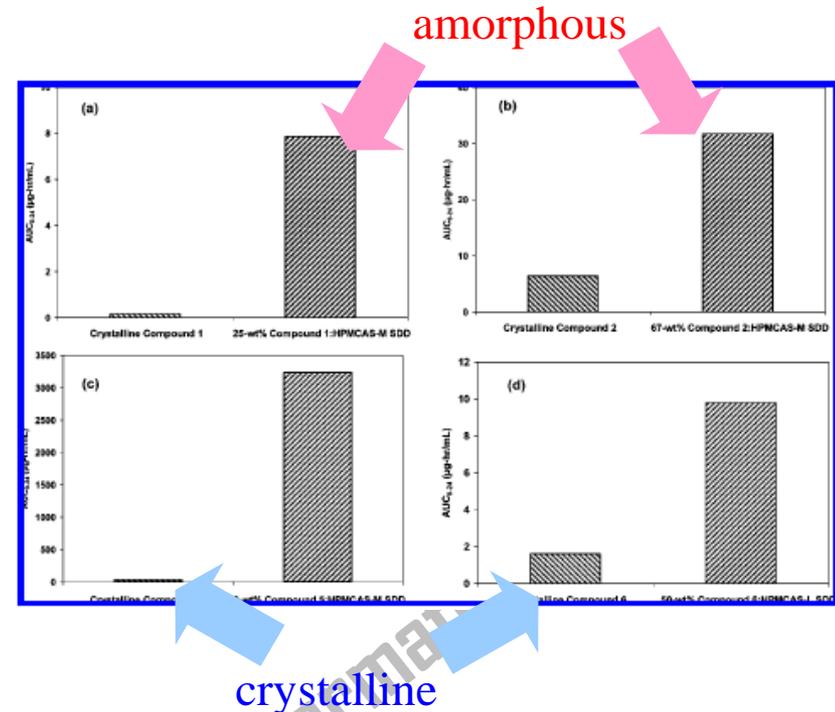


# 无定形可以提高药物溶出和生物利用度



**Dissolution Comparison:** amorphous (in PVP SD) vs crystalline Bicalutamide.

ANDREWS, GP et al., Journal of Pharmaceutical Sciences, Vol. 99, 1322–1335 (2010)



**Comparison of canine in vivo exposure** for crystalline and amorphous drug (in HPMCAS SDDs)

Friesen, DT et al, MOLECULAR PHARMACEUTICS VOL. 5, NO. 6, 1003–1019 (2008)

**Bioavailability:** fraction of administered dose of a drug that reaches systemic circulation



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# 无定形可以规避创新药对晶型专利的保护，提早将仿制药推向市场

RELATED KEYWORDS: Ranbaxy Laboratories

缬更昔洛韦

## Ranbaxy settles patent dispute with Roche

ET Bureau, Aug 26, 2010, 01:51am IST

Tags: Valcyte | Roche

NEW DELHI: Daiichi Sankyo-owned Ranbaxy Laboratories has settled its patent dispute with Swiss major Roche for the latter's drug valganciclovir sold under the brand Valcyte. The settlement was disclosed in a filing to a US appeals court in Washington so that it can be dismissed at the lower court. The request was granted on Tuesday, news agency Bloomberg said.

The terms of the settlement are not known. The Ranbaxy spokesman declined to comment while an email sent to the Roche spokeswoman in Basel, Switzerland remained unanswered. Ranbaxy is the first generic company to challenge valganciclovir's patent in the US. This means Ranbaxy can launch its generic version before valganciclovir's patent expires in 2015 with a 180-days marketing exclusivity if it succeeds in its patent dispute.

**180-days marketing exclusivity**



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# 市场上的无定形药物

## • Marketed Drug Products

- **Accolate** (zafirlukast, 安可来, ASTRAZENECA )
- **Accupril** (quinapril hydrochloride, 喹那普利 )
- **Ceftin** (cefuroxime axetil, 头孢呋辛酯, GSK)
- **Humulin\*** (insulin, 优泌林 )
- **Rezulin** (troglitazone, 曲格列酮 )
- **Sporanox** (itraconazole, 斯皮仁诺, J&J )



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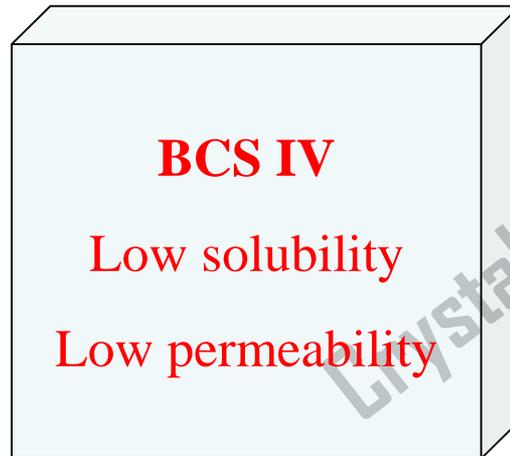
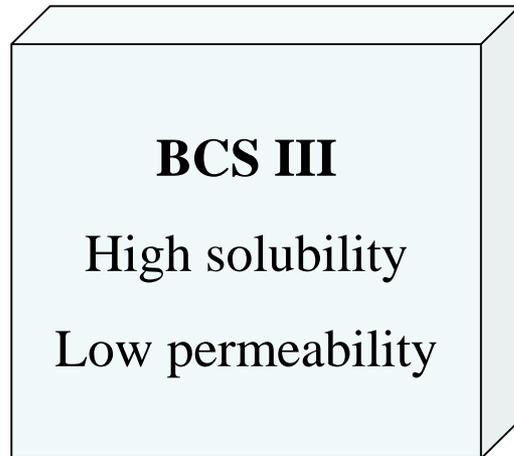
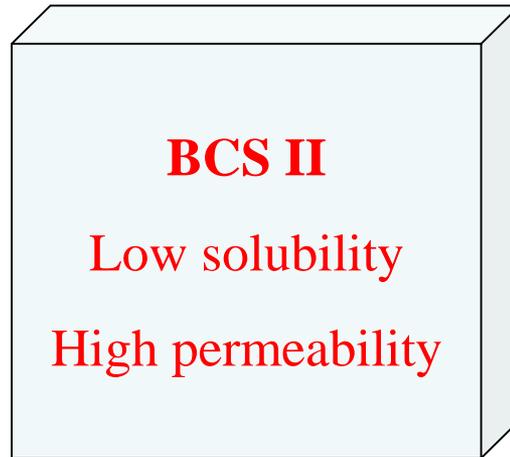
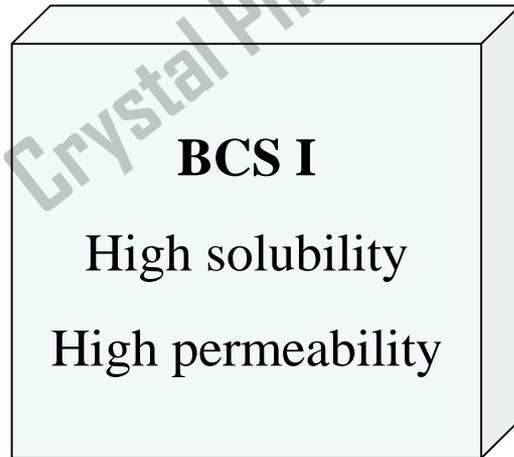


\*Partially amorphous designed for rapid onset + extended release (crystalline fraction)



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# 药物分子的BCS 分类



Dose number:

$$D_0 = (M_0/V_0)/C_s$$

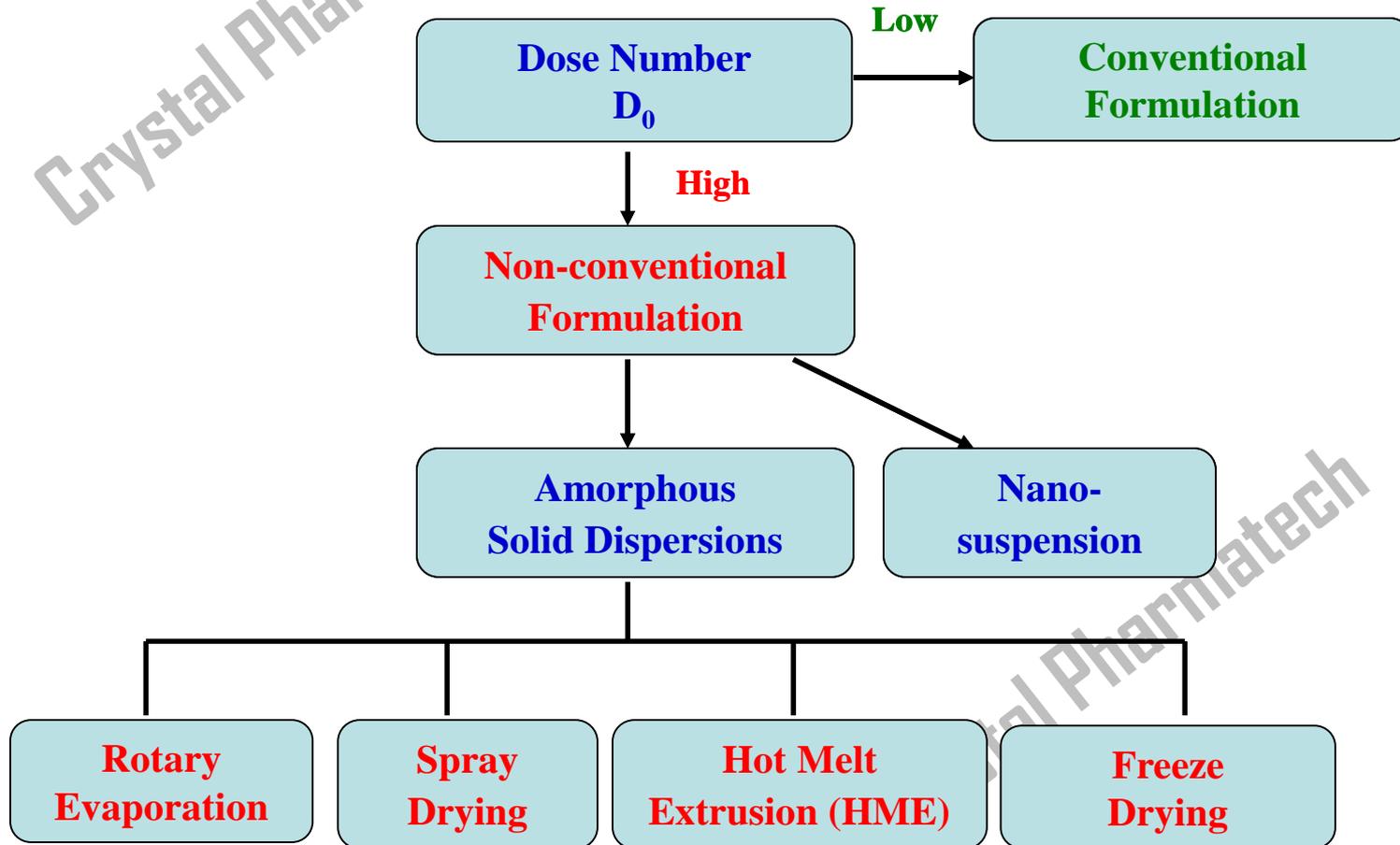
Dose volume:

$$V_d = M_0/C_s$$

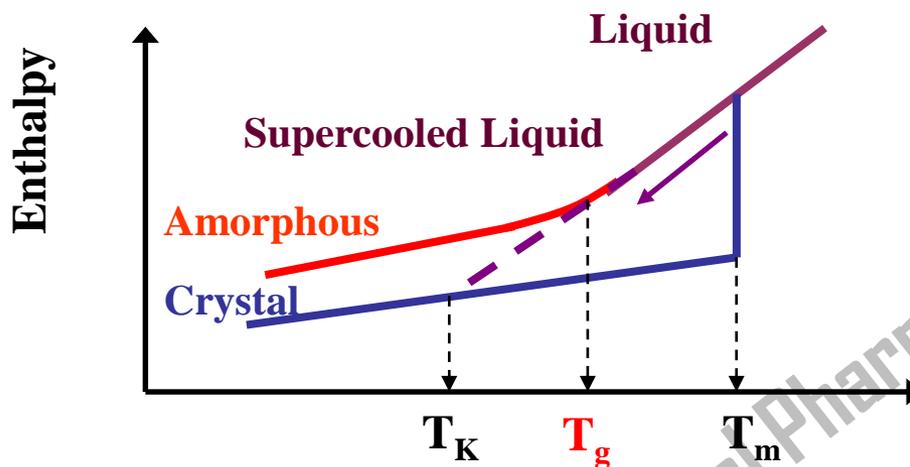
Large  $D_0$ , More challenging



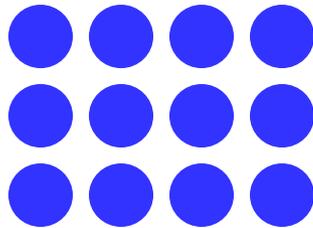
# 药物研发中不同制剂方法的选择



# 无定形的基础知识

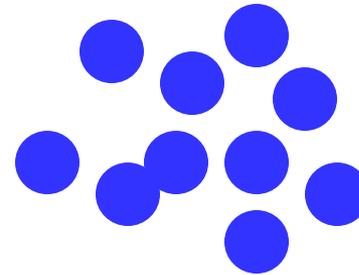


# 晶相和无定形



Molecules in a ordered, repeating pattern

- Equilibrium state
- More stable
- Lower solubility
- Lower dissolution rate

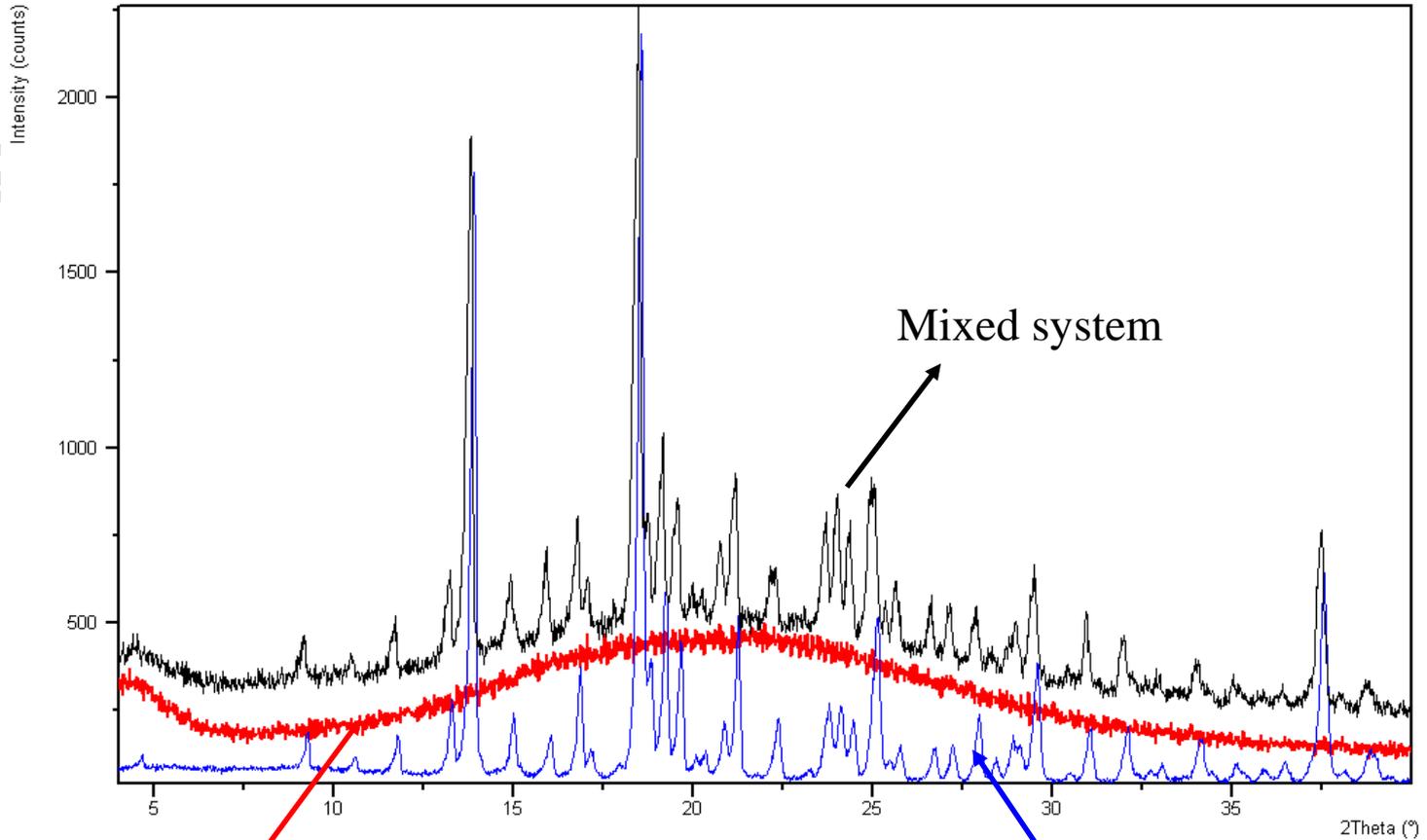


Molecules in a random arrangement with some local order

- Non-equilibrium state
- Typically less stable
- Higher apparent solubility
- Higher dissolution rate



# 晶型 vs. 无定形: XRPD



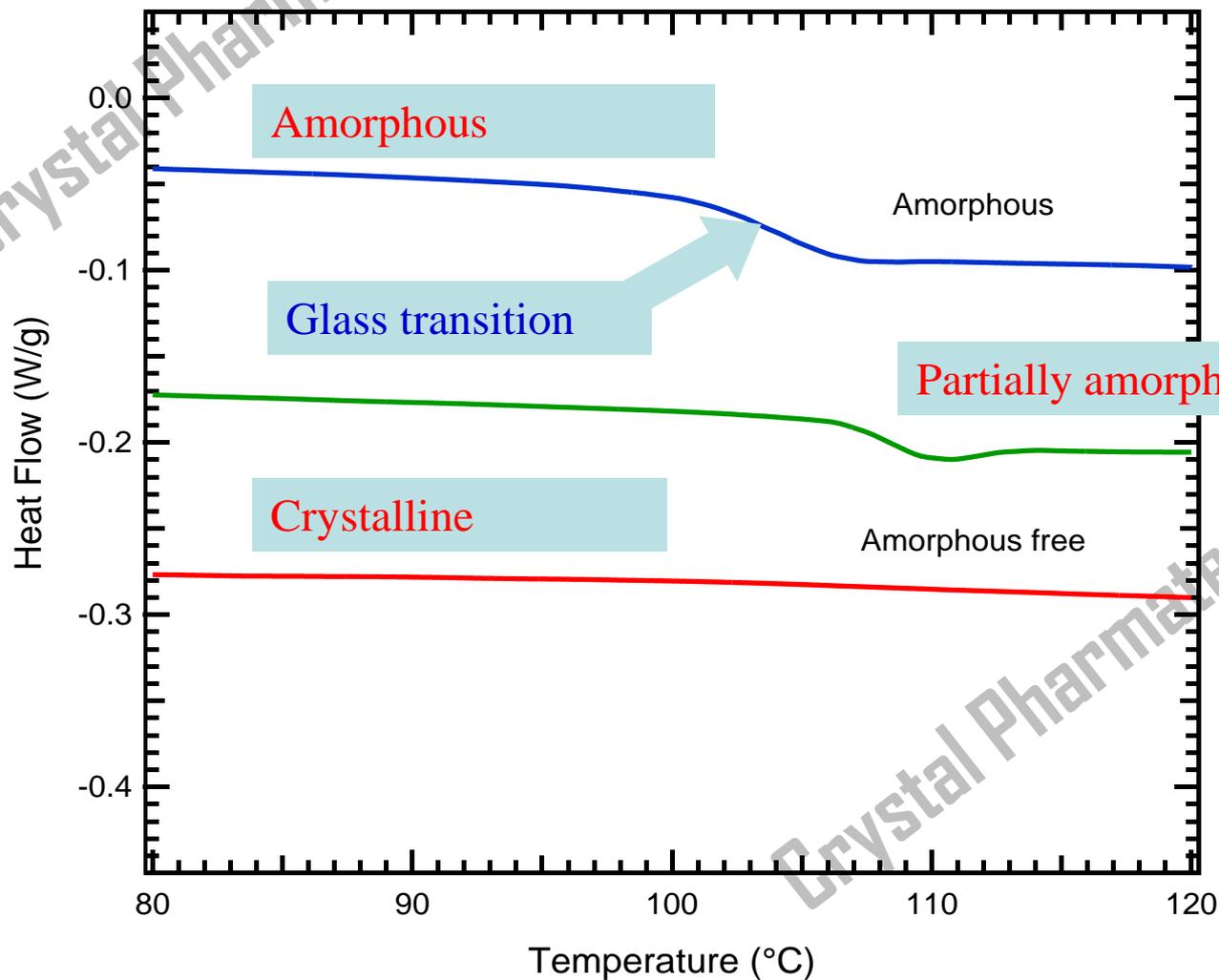
100% amorphous

100% crystalline

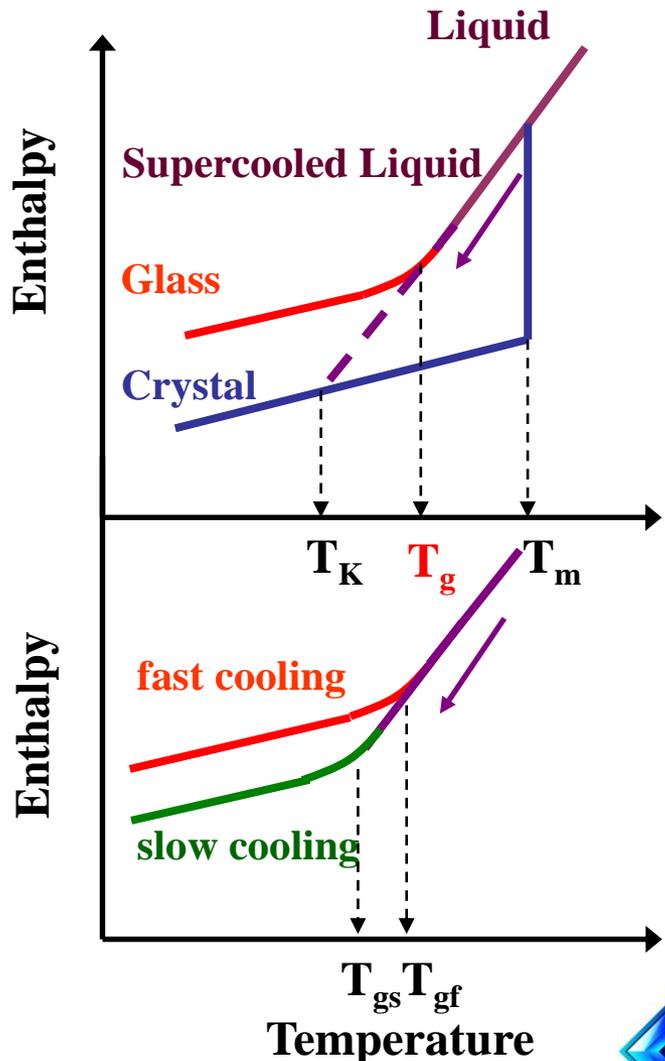


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# 晶型 vs. 无定形: DSC/mDSC



# 超冷却液体和无定形固体



Upon cooling, molecular motions in liquids slow down. If crystallization is avoided below its freezing point, the liquid will exist in a supercooled state. Upon further cooling, the molecules in the supercooled liquid will rearrange so slowly that they appear 'frozen' on the laboratory timescale. This falling out of equilibrium occurs across a narrow transformation range where the characteristic relaxation time is of the order of 100 seconds. This transformation is normally referred as the glass transition. The slower a liquid is cooled, the colder it can become before falling out of equilibrium.

$T_g$ : glass transition temperature

$T_k$ : Kauzmann temperature

$T_m$ : melting point

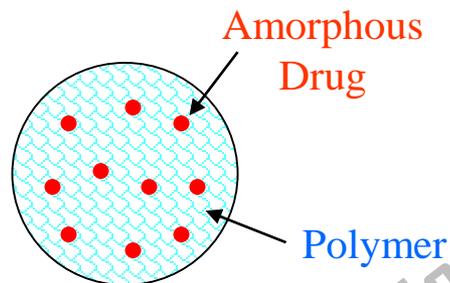


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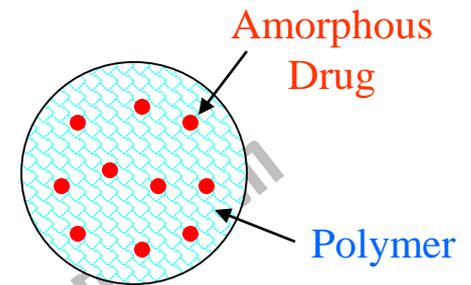
*Debenedetti, P. G.; Stillinger, F. H.  
Nature, 2001, 410, 259-267*

# 无定形固态分散物制剂



# 什么是无定形固态分散物？

- A single phase amorphous solid solution of drug, polymer, optional surfactants, etc. as determined by common analytical techniques such as XRPD, DSC, ssNMR.



- Methods to obtain solid dispersions include: spray drying, hot melt extrusion and freeze drying, etc.



# 无定形固态分散物：药物释放

Particle Sizes



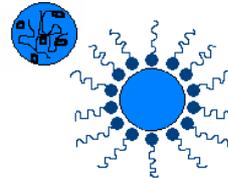
Solid dispersion

Dissolution



Larger particles  
>500 nm

-500 nm



- Amorphous nanoparticles
- Bile salt micelles
- etc.

-5 nm



“Molecularly dispersed  
supersaturation”

(Recrystallization)



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# 无定形分散物中辅料的选择

- Polymers
  - Carriers for the drug
  - Inhibit crystallization both in dosage form and in-vivo
- Surfactants
  - Solubilizers or emulsifying agents
- Organic solvents
  - Needed to dissolve all components



# 无定形分散物中高分子的选择

- Increase aqueous solubility
  - Use water soluble macromolecules
  - Use amorphous polymers
- Improve physical stability (inhibit crystallization)
  - Use macromolecules
    - Reduced mobility (long relaxation times)
    - Produce single phase solid solutions with drugs
    - Use polymers with high  $T_g$
    - Frustration against crystallization



# 无定形分散物中高分子的选择

Table 1: Properties of Polymers Commonly Used in ASDs [7]

Polymer	Tg (°C)	Solvent Solubility	Hygroscopicity	Amenable Methods of Manufacture
Copovidone	106	Dichloromethane Ethanol Methanol Water Acetone	<10% @ 50% RH	Rotary Evaporation Spray Drying Hot Melt Extrusion
Polyvinyl caprolactam-polyvinyl acetate-polyethyleneglycol copolymer	70	Water Ethanol Methanol Acetone	~5% @ 50% RH	Rotary Evaporation Spray Drying Hot Melt Extrusion
PVP	130 (K17) 168 (K30)	Chloroform Ethanol Methanol Water Acetone	~15% @ 50%RH	Rotary Evaporation Spray Drying Hot Melt Extrusion
HPMC	170	Cold Water Dichloromethane: Ethanol Dichloromethane: Methanol Water: Alcohol	<10% @ 50% RH	Spray Drying
HPMC P	133 - 137	Acetone: Methanol Acetone: Ethanol Methanol: Dichloromethane	2 - 5% @ 50%RH	Spray Drying
HPMC AS	110 - 130	Acetone* Ethanol:Dichloromethane* *clear or turbid viscous solution	~3% @ 50%RH	Spray Drying
Methacrylate/methacrylic acid copolymer	110 - 150	Ethanol, Methanol, Acetone, Acetone with 3% water	<5% @ 50% RH	Rotary Evaporation, Spray Drying, Hot Melt Extrusion

*Reference: Amorphous solid dispersions as enabling formulations for discovery and early development, American Pharmaceutical Review, Jan/Feb 2011*

# 无定形分散物中表面活性剂的选择

**Table 2. Properties of Surfactants Commonly Used in ASDs [7]**

Surfactant	HLB	T <sub>m</sub> (°C)	Solvent Solubility
Vitamin E Polyethylene Glycol Succinate	13.2	37 - 41	Water Acetone
Sorbitan Monostearate – 60/80	4.3	53 - 57	Most organic solvents
Polysorbate 20	16.7	n/a	Ethanol Water
Polysorbate 80	15.0	n/a	Ethanol Water
Polyoxyl 40 Hydrogenated Castor Oil	14 - 16	30	Chloroform Ethanol Acetone Water

*Reference: Amorphous solid dispersions as enabling formulations for discovery and early development, American Pharmaceutical Review, Jan/Feb 2011*



# 无定形分散物工艺中溶剂的选择

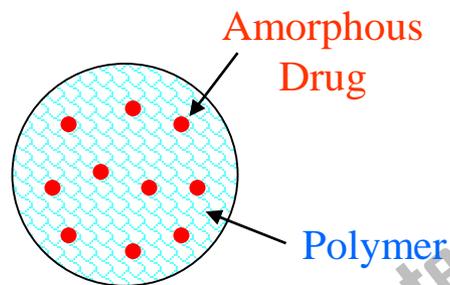
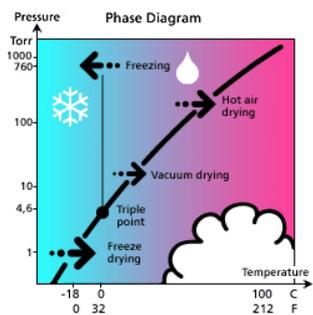
**Table 3. Properties of Organic Solvents Commonly Used for Preparation of ASDs [8]**

Solvent	Boiling Point (°C)	Flash Point (°C)	ICH Class
Ethanol	78	13	II
Methanol	65	12	III
Acetone	57	17	III
IPA	83	12	III
Tetrahydrofuran	66	-14	III
Dichloromethane	40	None	II

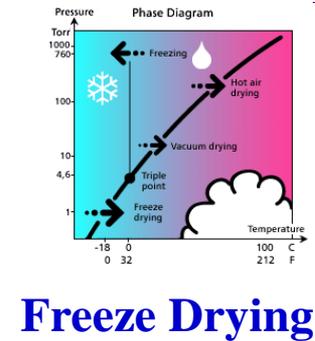
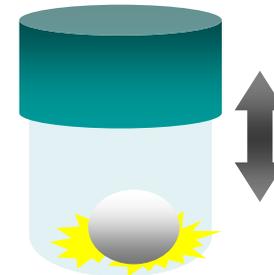
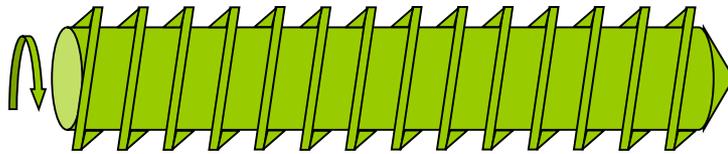
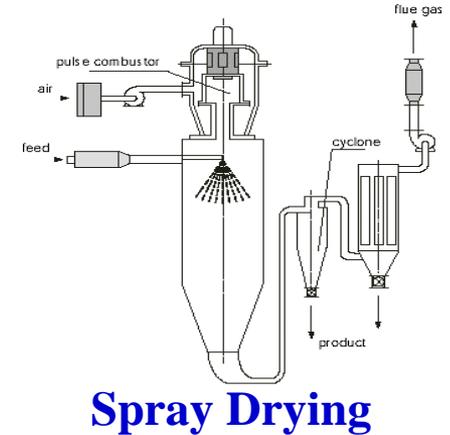
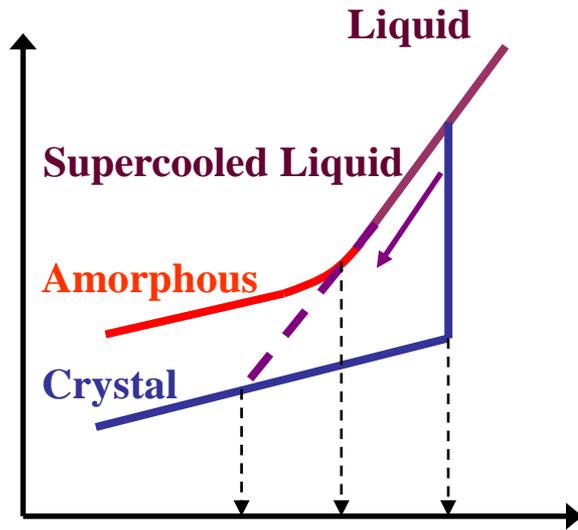
*Reference: Amorphous solid dispersions as enabling formulations for discovery and early development, American Pharmaceutical Review, Jan/Feb 2011*



# 无定形固态分散物制剂的制备



# 怎样得到无定形？



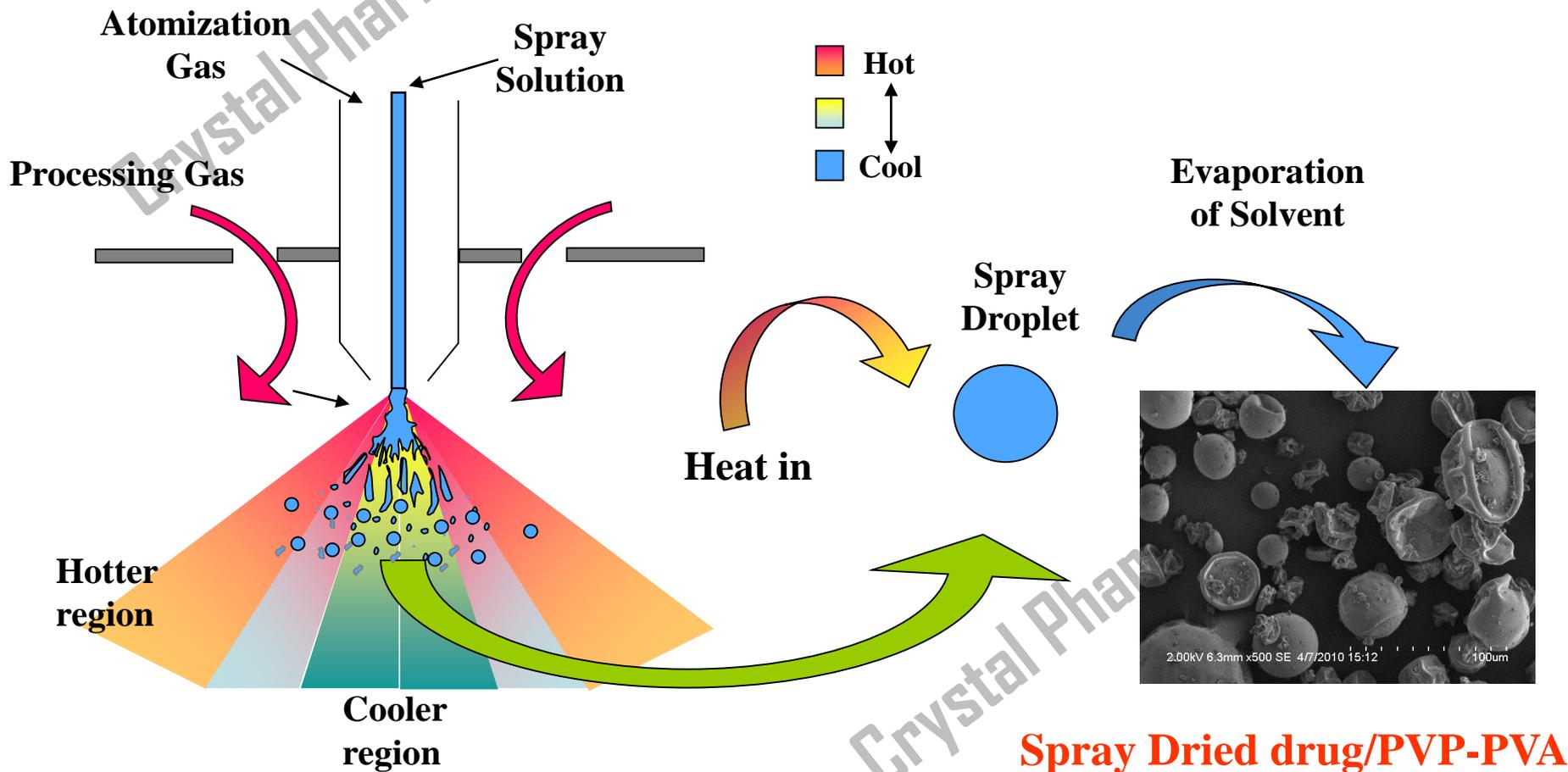
Hot Melt Extrusion

Ball Milling

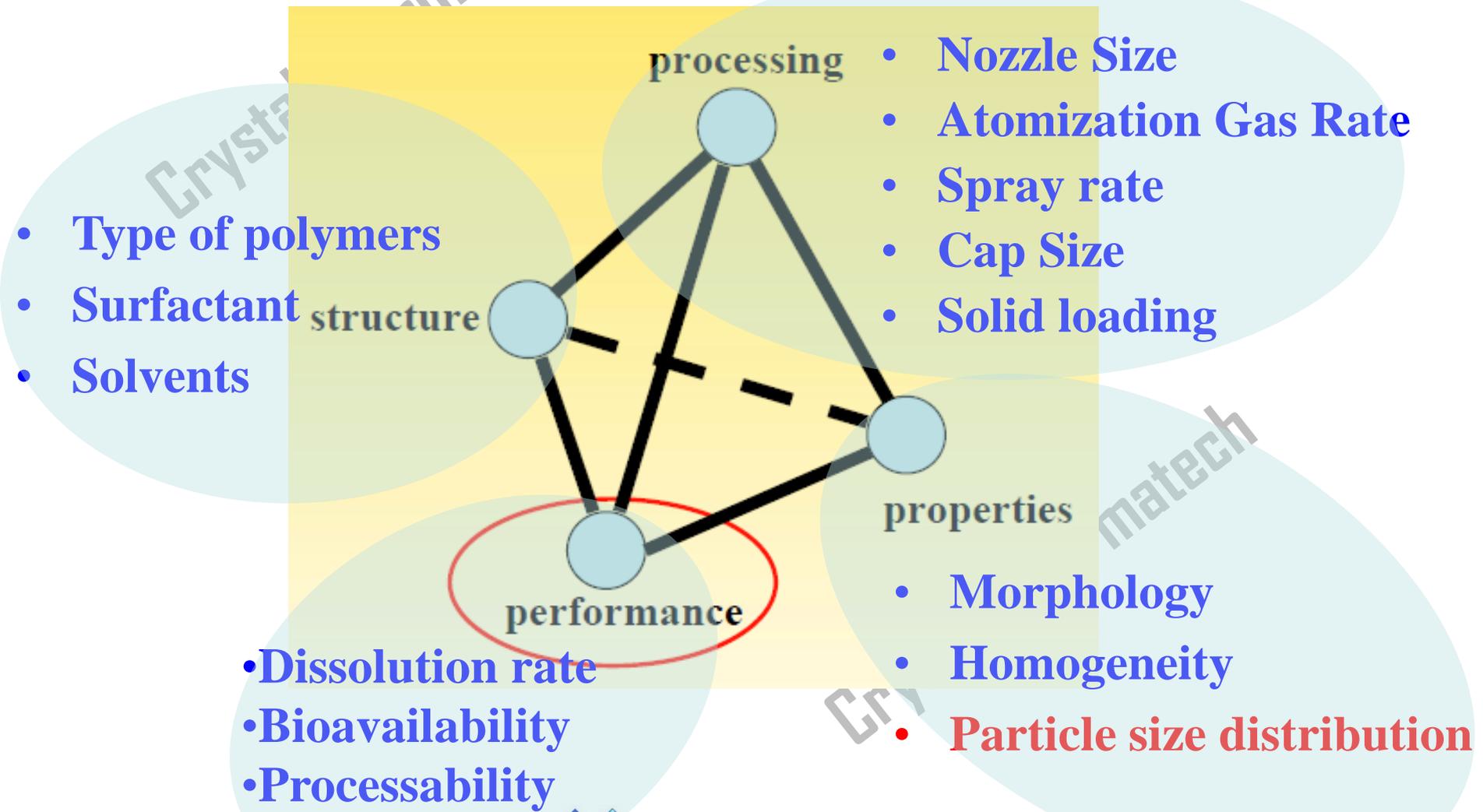
Freeze Drying



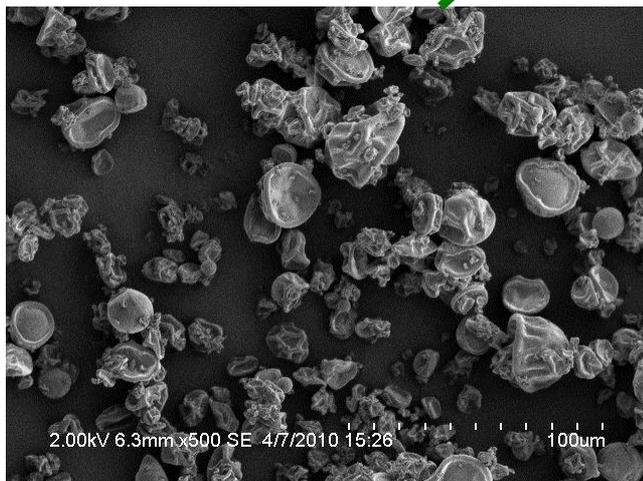
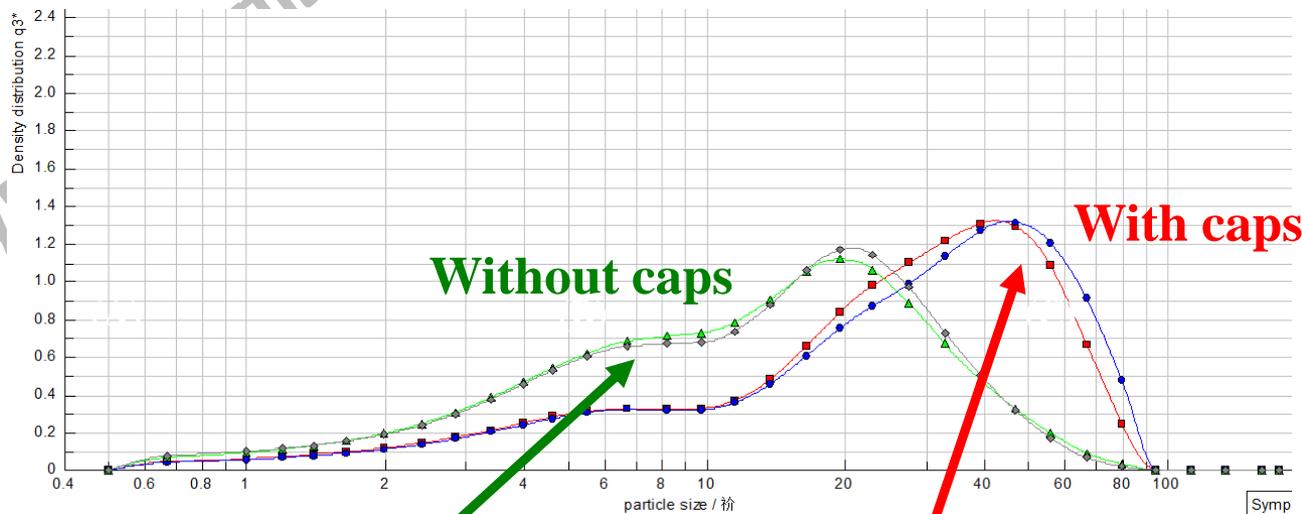
# 喷雾干燥法



# 喷雾干燥工艺参数对产品质量的影响

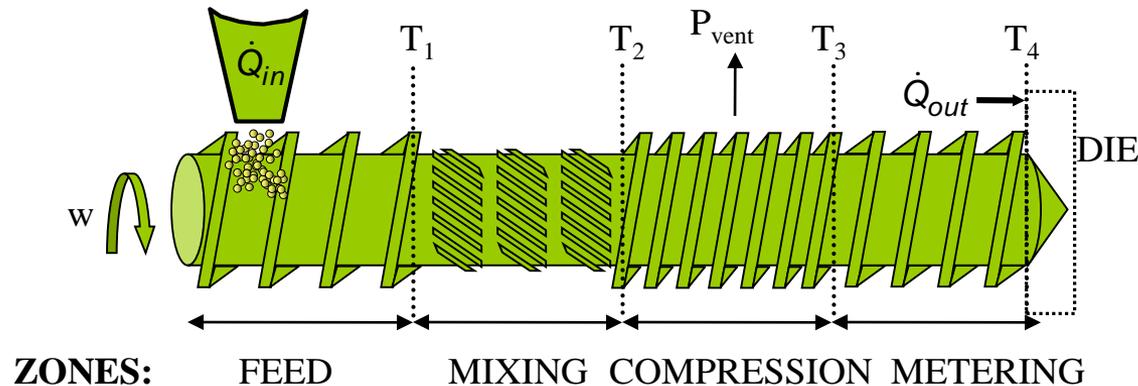


# 喷雾干燥工艺参数对产品质量的影响



# 熔融挤出法

## Modern Engineering Polymer Extrusion



**robust platform**

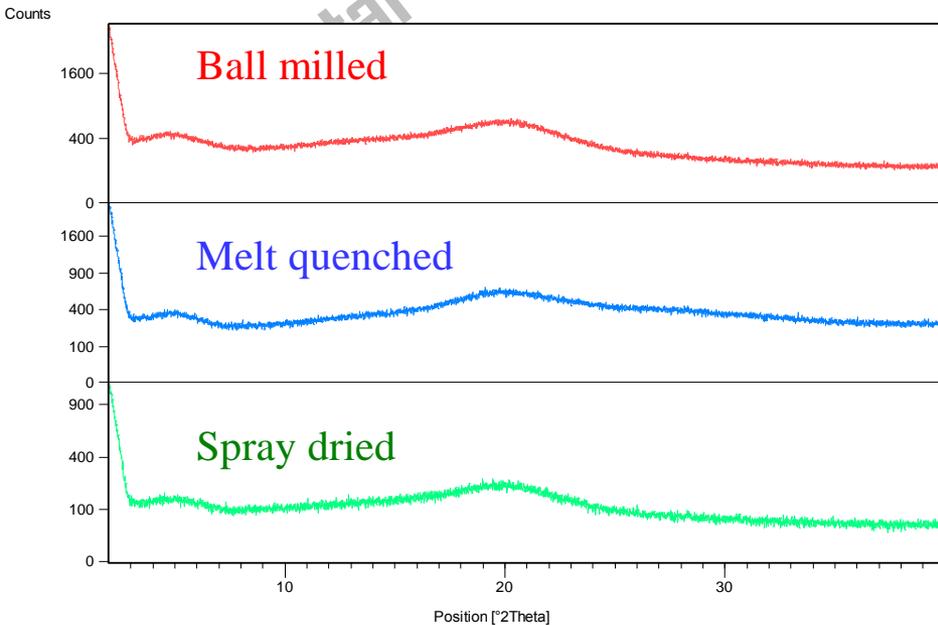
- Zone Temp control
  - Deliberate Mixing
  - Compression
  - Venting
  - Addition of liquid to granulate via side ports
- ⇒



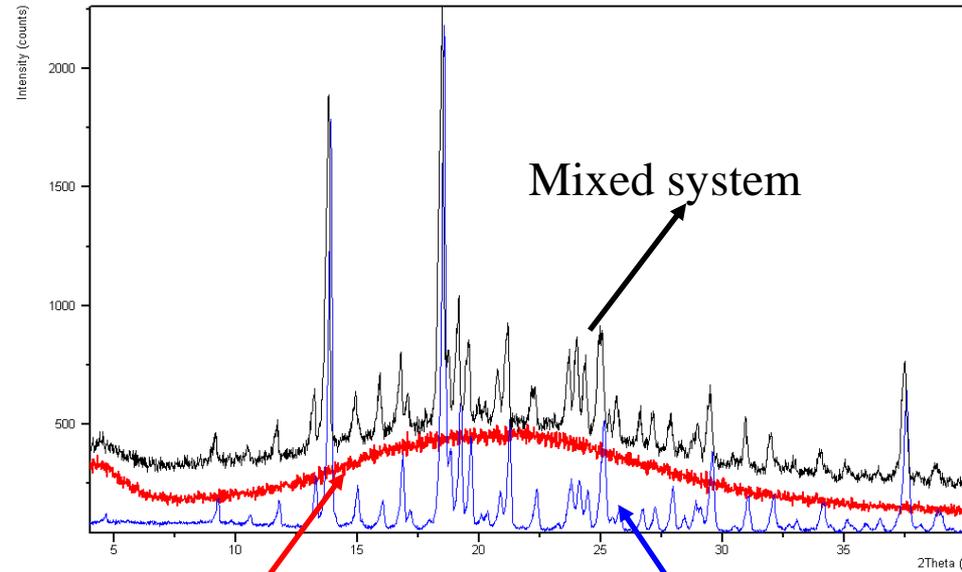
# 无定形的检测和表征



# 无定形的检测和表征: XRPD



Amorphous halo by XRPD: characteristic

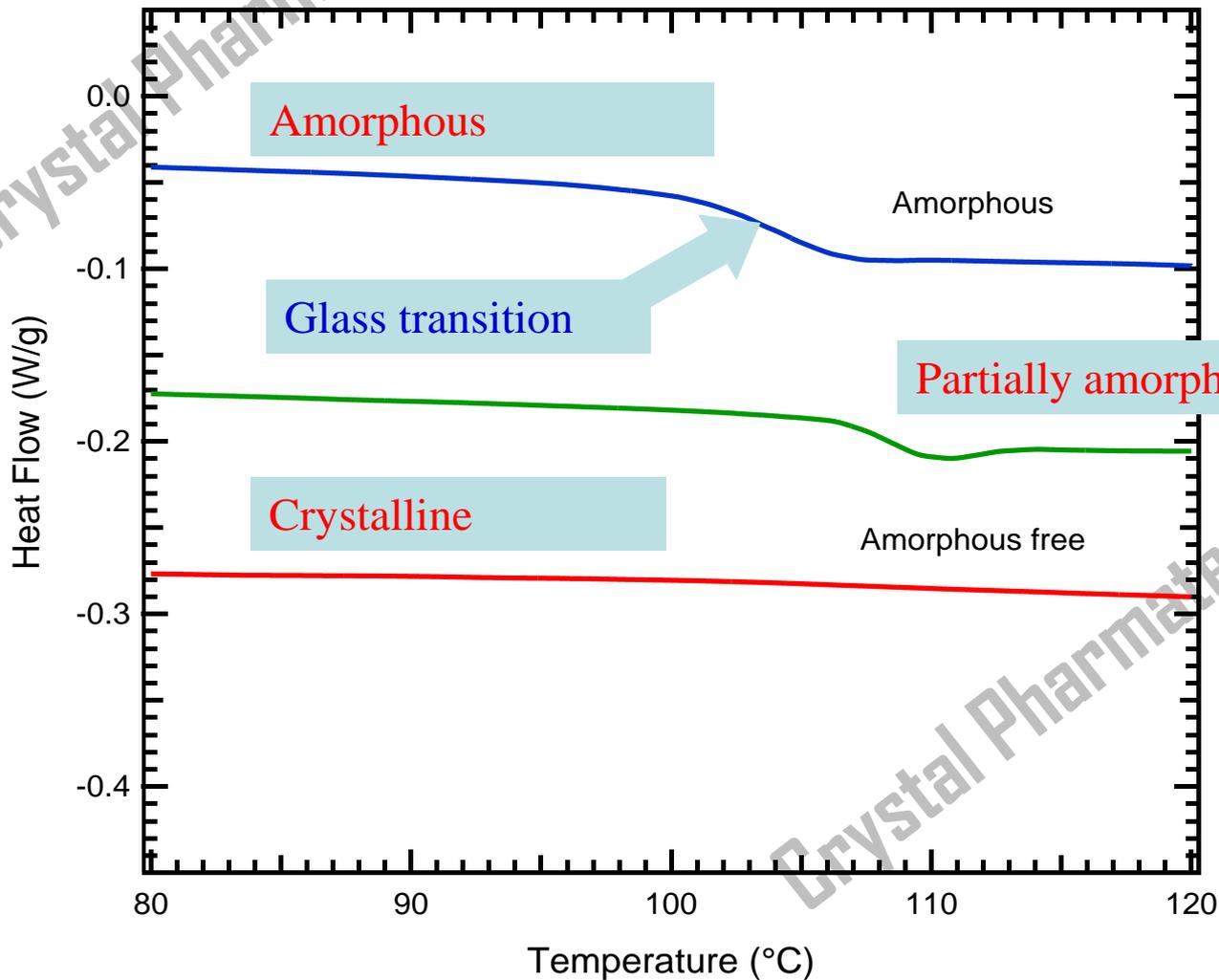


100% amorphous

100% crystalline



# 无定形的检测和表征: DSC/mDSC

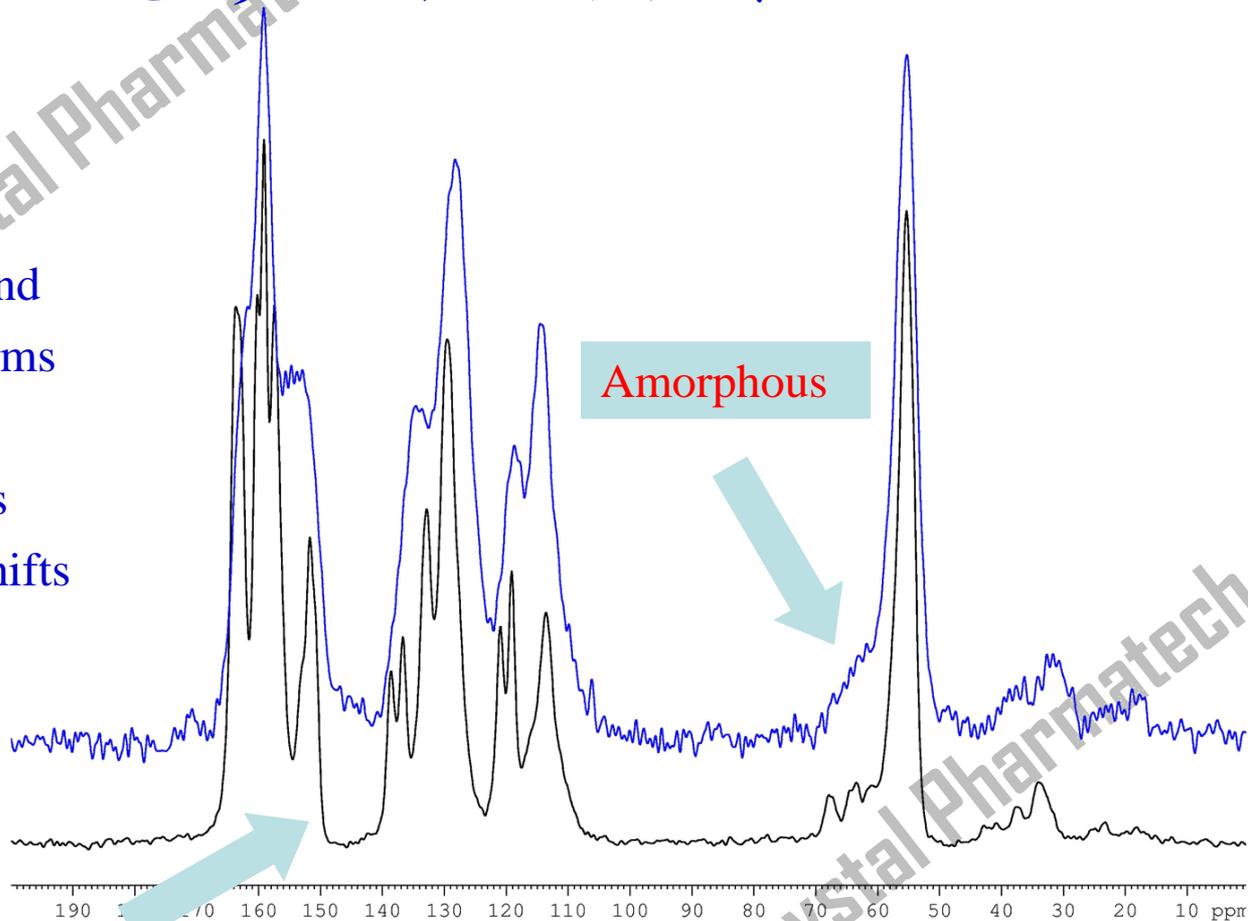


# 无定形的检测和表征: ssNMR

Amorphous and crystalline forms identifiable:

→ Linewidths

→ Solution shifts



Crystalline

Amorphous

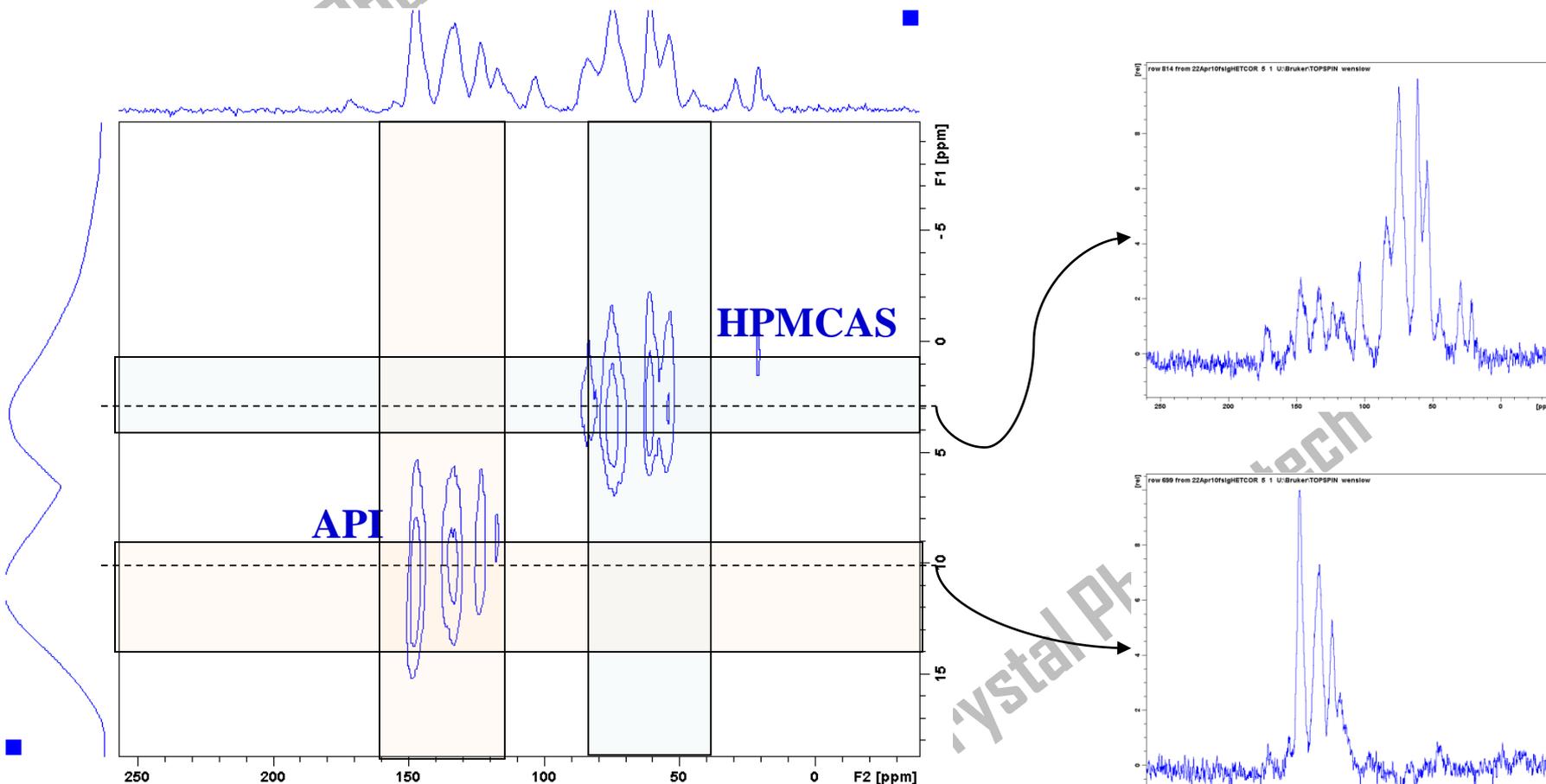


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# 固态核磁共振应用

## HETCOR- $^{13}\text{C}$ projections

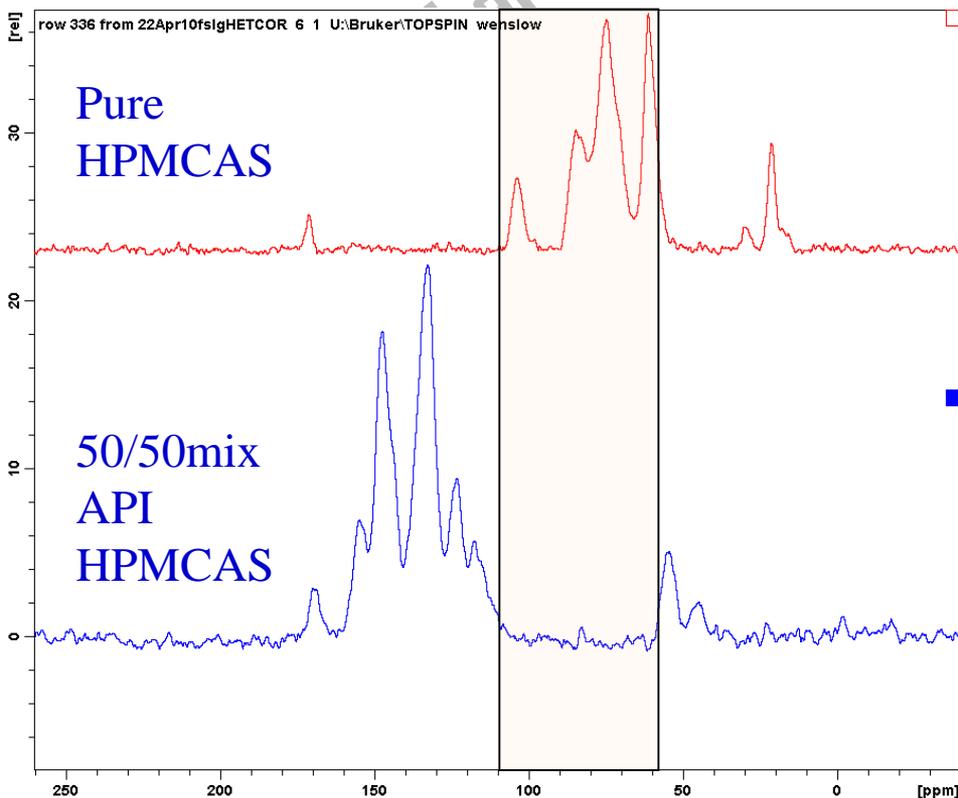


*No Molecular Level Mixing*

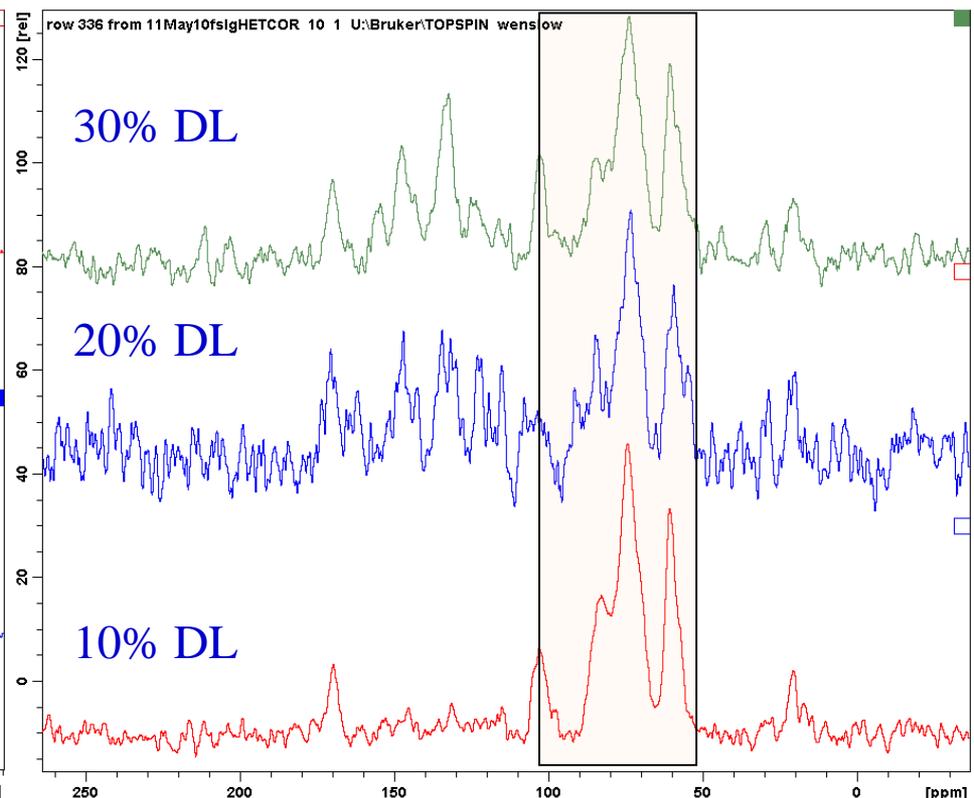


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# 固态核磁共振应用: 无定形固态分散物



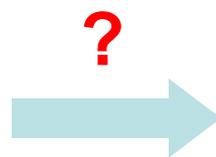
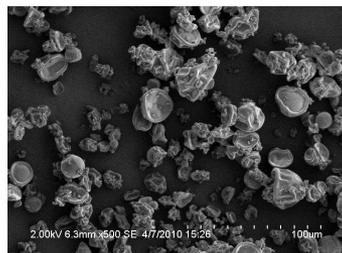
*No Molecular Level Mixing*



*Significant Molecular Level Mixing*

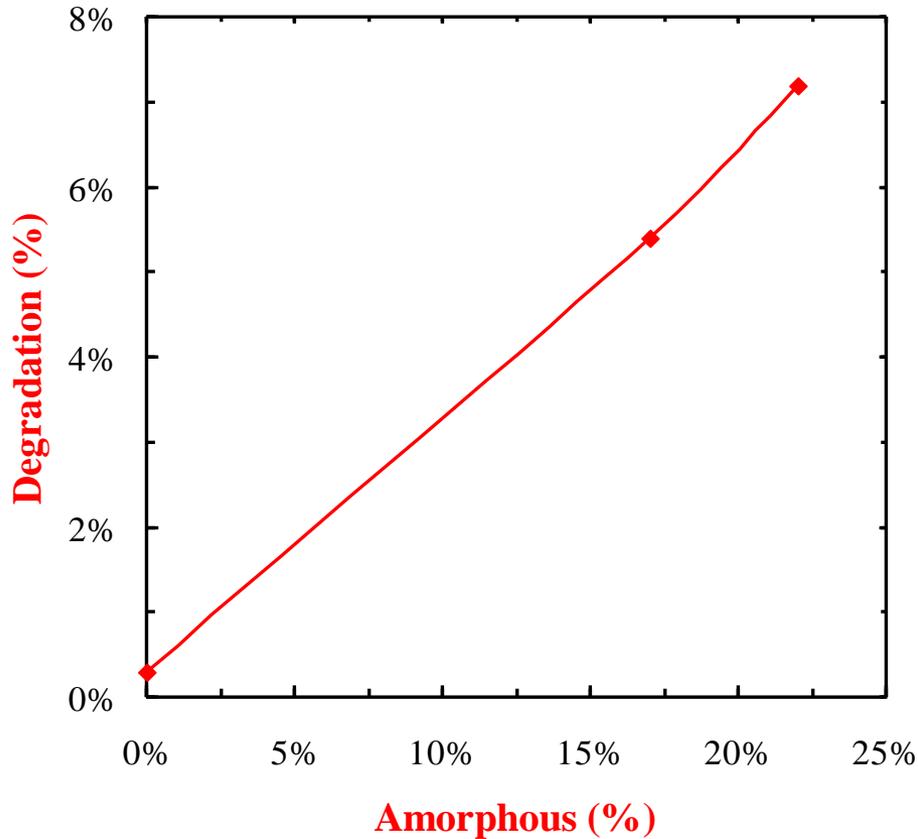


# 无定形的稳定性研究

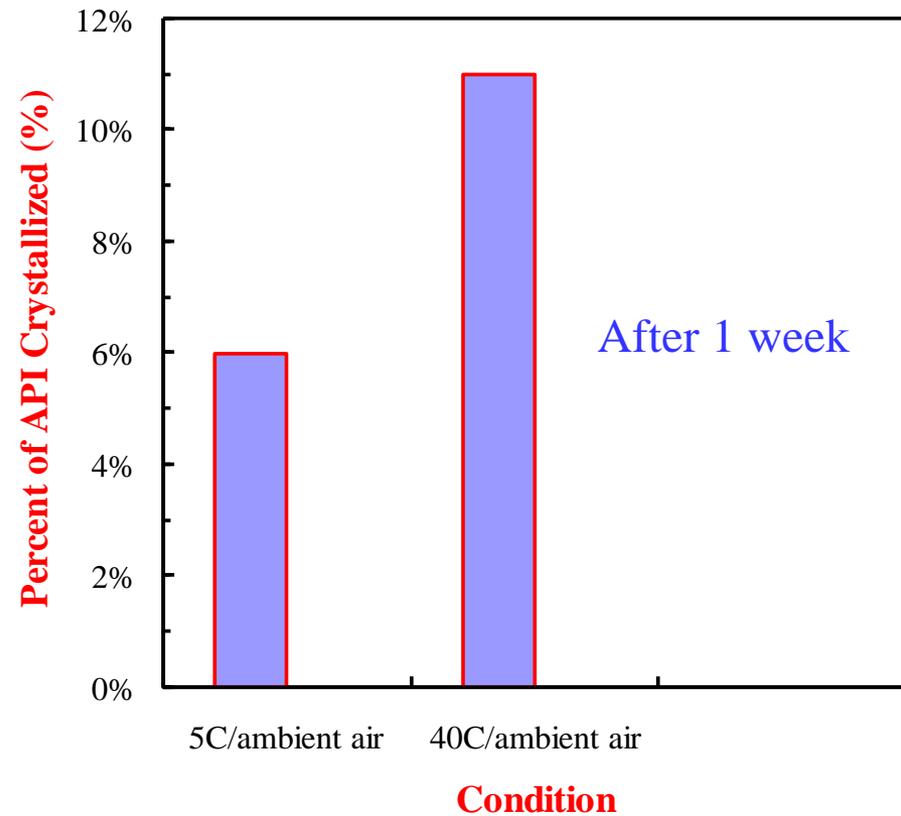


# 无定形原料药和制剂的稳定性

Effect on chemical stability: 40C/75%/2wk



physical stability: 40C/75%/2wk



20% API + 75% HPMC-ASHF + 5% Tween



# 无定形体系稳定性的影响因素

**The stability of an amorphous phase depends on**

(a) **Thermodynamic properties** of the amorphous phase, such as

- Configurational entropy ( $S_c$ )
- Excess enthalpy relative to a crystalline phase ( $\Delta H_{ex}$ )

(b) **Dynamic properties** of the amorphous phase, such as

- Structure relaxation (alpha-relaxation)
- Beta-relaxation process
- Activation energy barrier

(c) Other factors such as the presence of nuclei, additives, and thermal history (sample preparation)



# 无定形体系稳定性的重要物理参数

– Chemical/supramolecular structure

–  $T_g/T_k/T_0$

– Structure relaxation

– Energy landscape perspective

– Configurational entropy

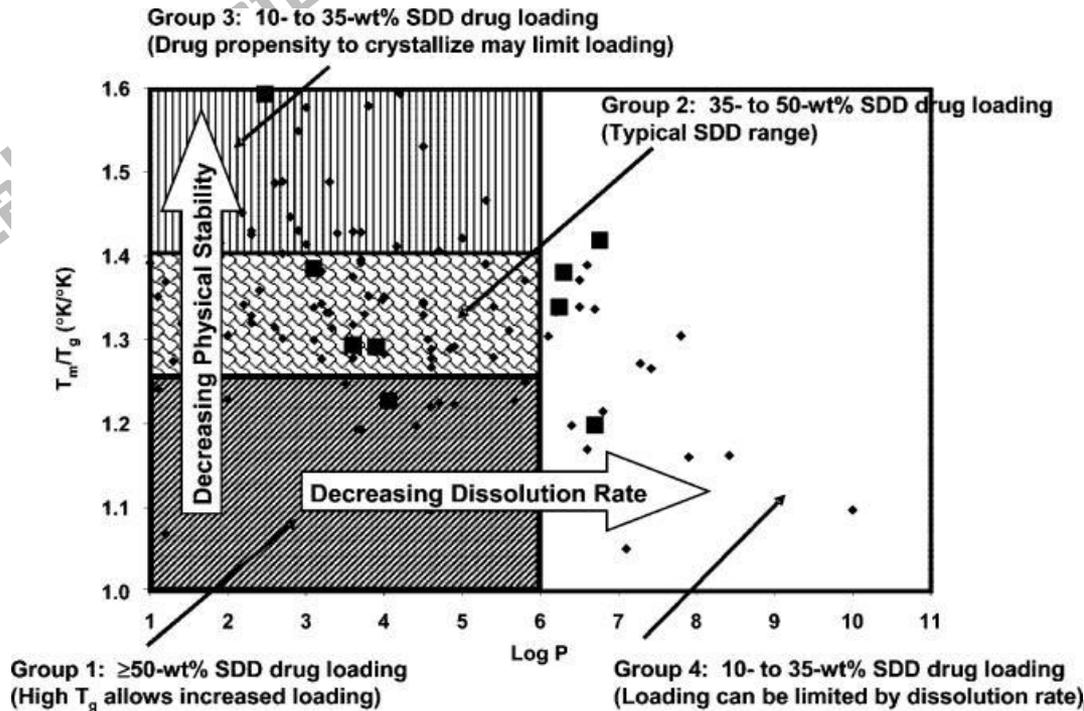
– Interfacial energy

– Beta-relaxation process

– Activation energy barrier to nucleation



# 无定形体系的稳定性

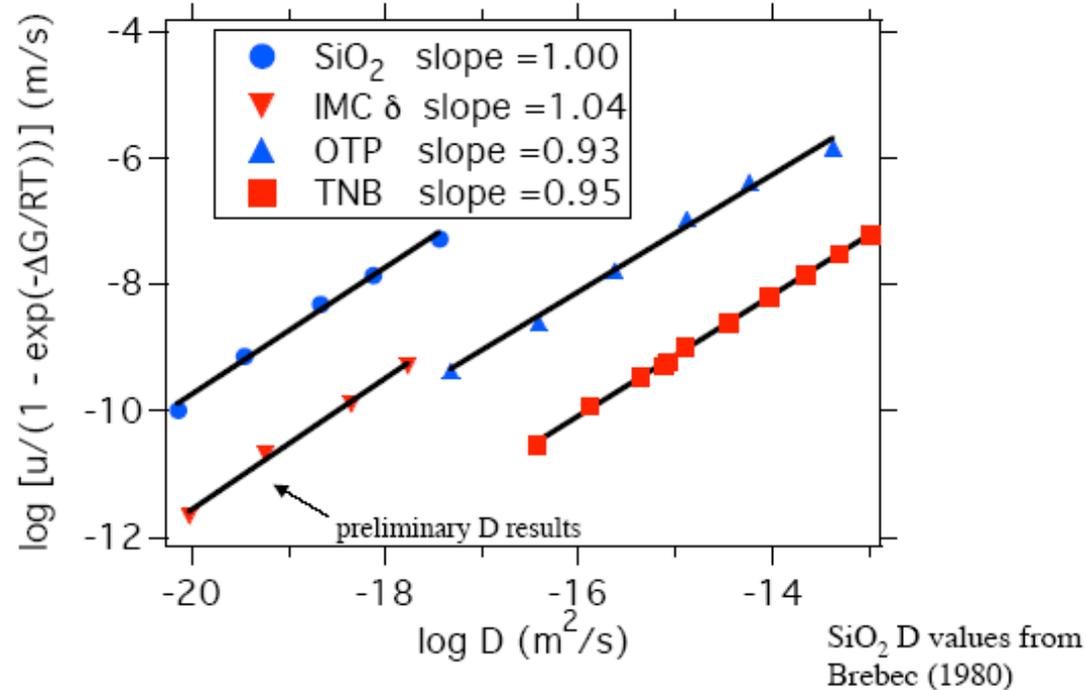
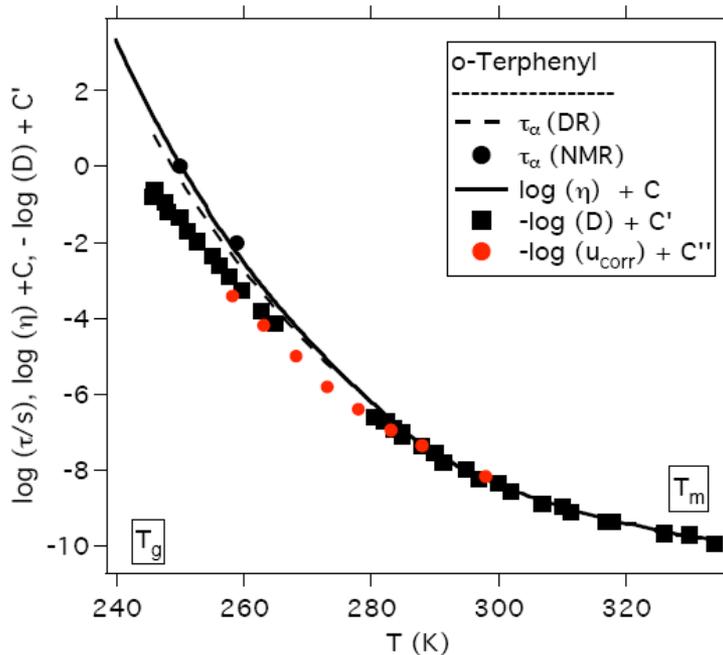


- Compounds with high  $T_m$  have high tendency to crystallize
- Compounds with high  $T_g$  have high kinetic barrier for mobility
- Compounds with high  $\log P$  ( $>4$ ) could have wetting problems in aqueous media



# 案例：无定形体系中晶体生长速率

Crystal growth rates could depend on  $D$ ,  $\eta$

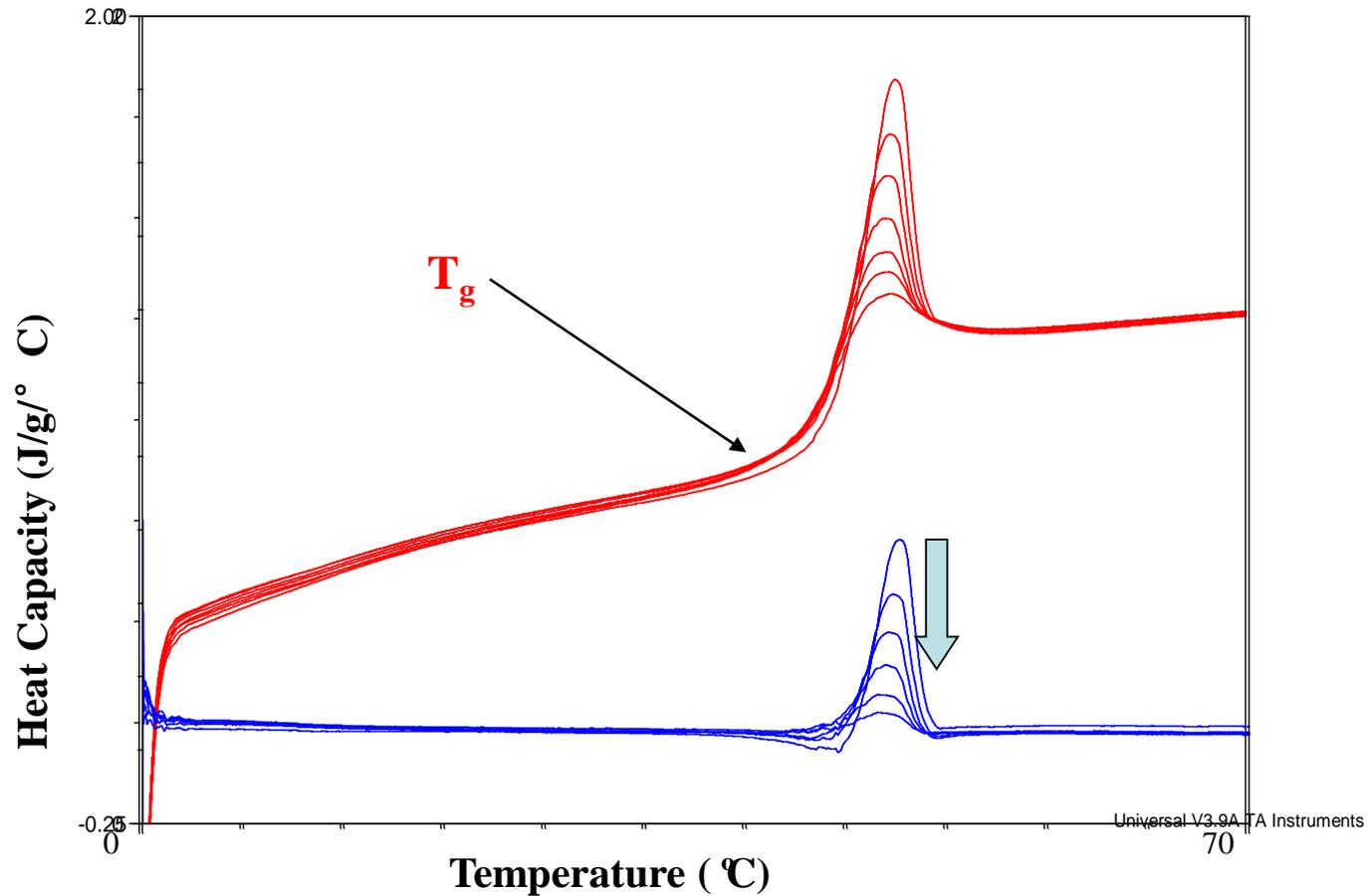


$u$  follows  $D$  better than  $\eta$

$D$  reasonably predicts  $u$  for many systems



# 无定形结构弛豫研究：DSC



# 无定形结构弛豫研究：DEA

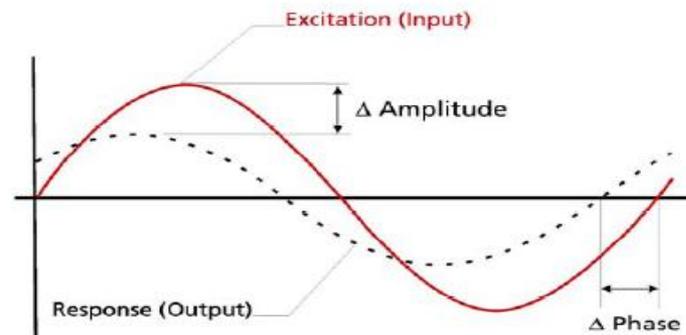
## Relative dielectric constant

$$\varepsilon^* = \varepsilon' - i\varepsilon''$$

**$\varepsilon'$ : permittivity**, a measure of the alignment and number of dipolar groups

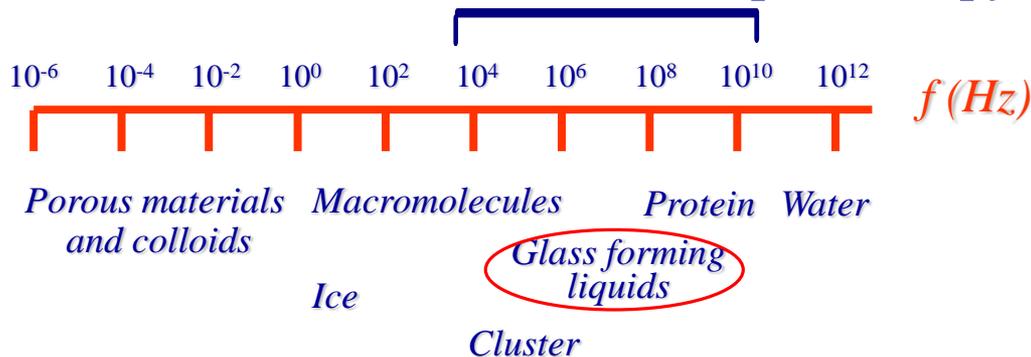
**$\varepsilon''$ : loss factor**, a measure of total energy loss due to the work performed aligning dipoles and moving ions

## Dielectric Relaxation

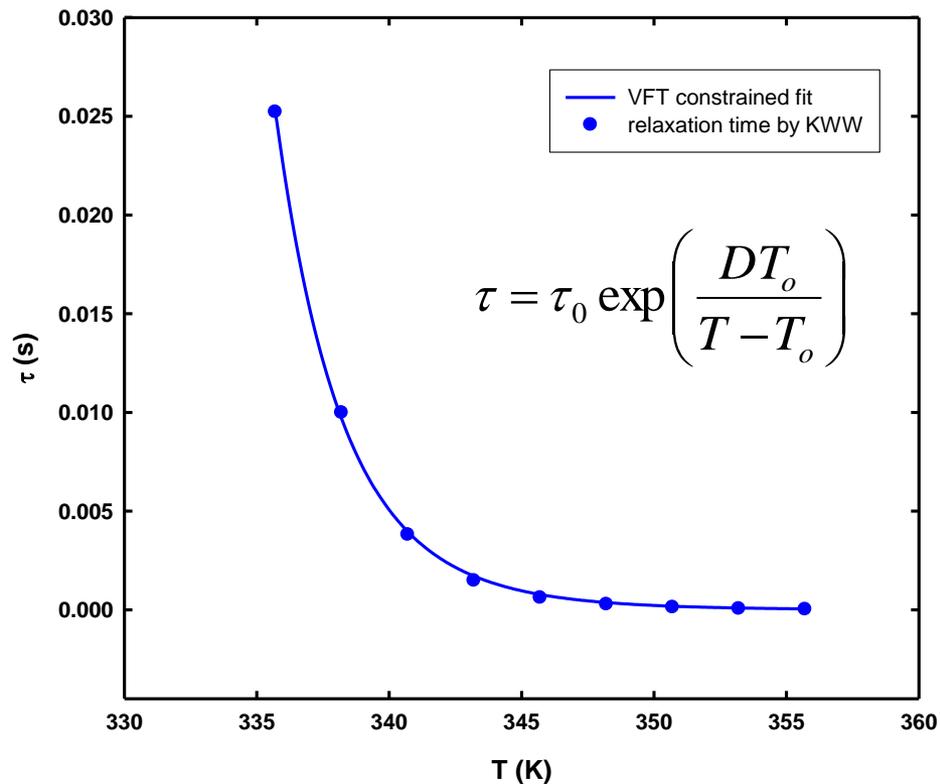
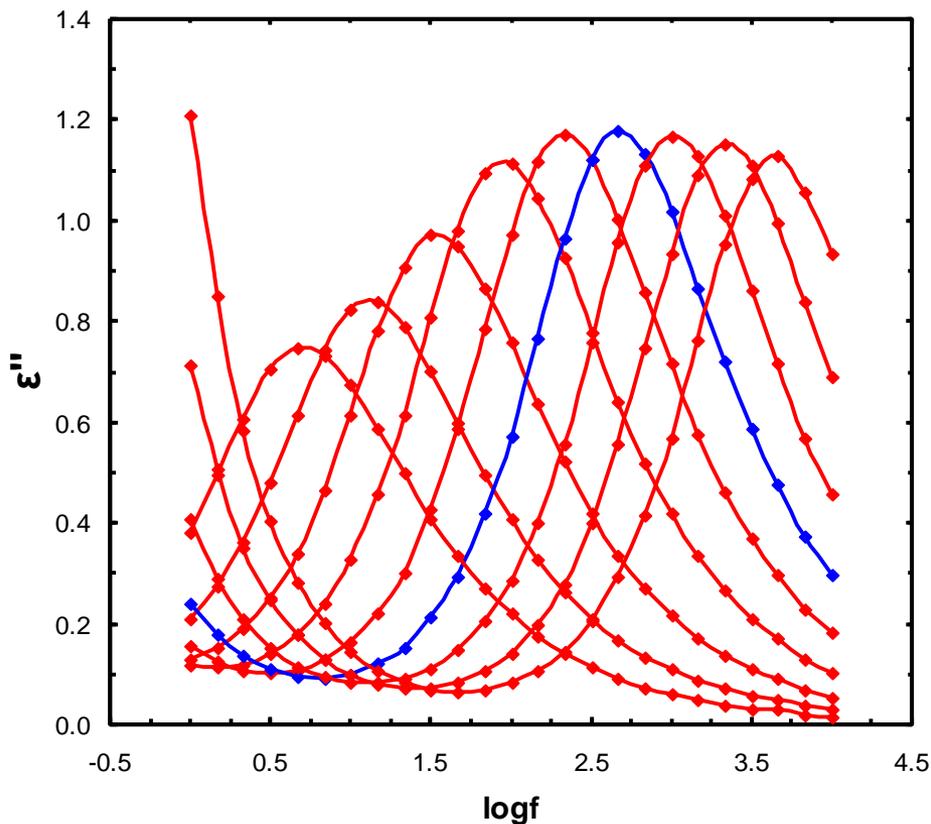


## Dielectric Spectroscopy

### Time Domain Dielectric Spectroscopy



# 无定形药物的弛豫现象研究: DEA

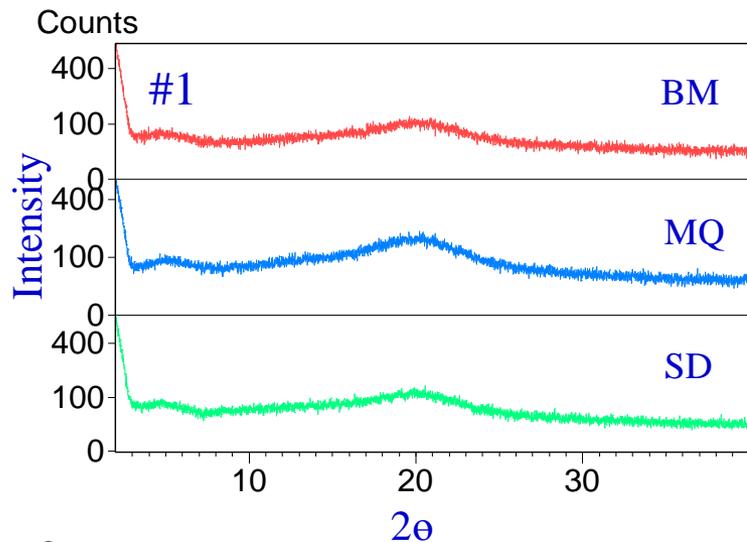


$$D = 7.4965, T_0 = 267.3$$

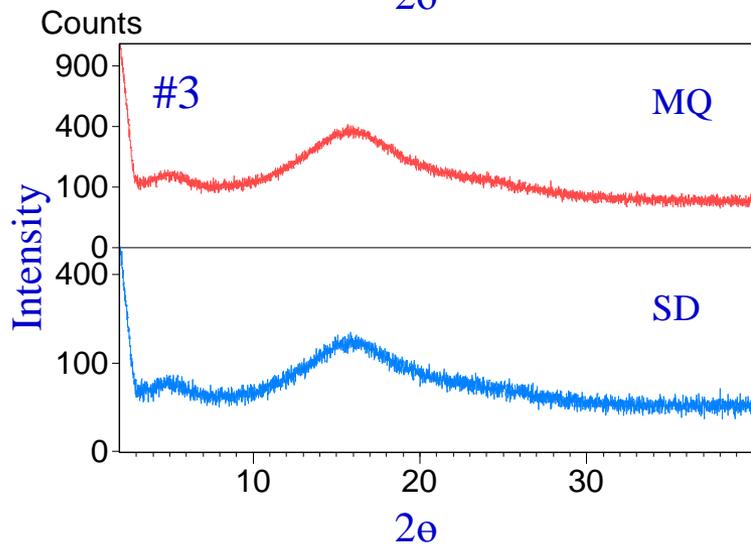
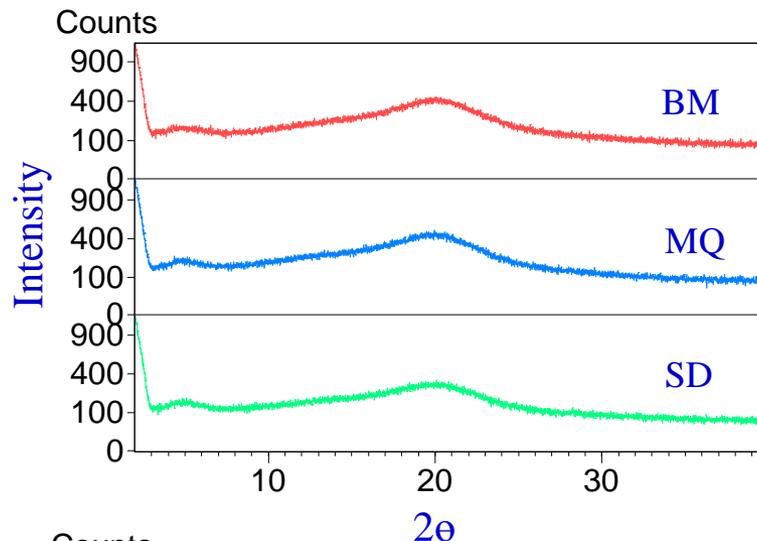
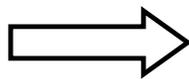
$$T_g = 321.7\text{K} (48.5^\circ \text{C})$$



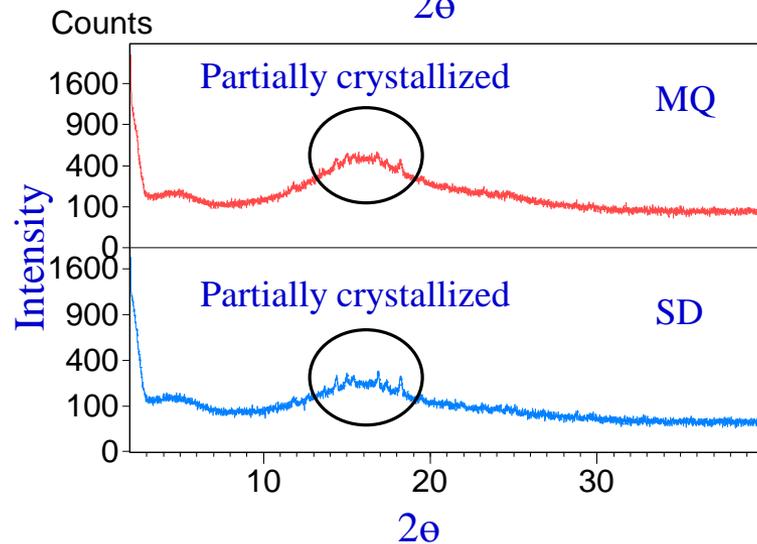
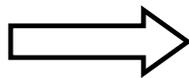
# 无定形药物物理稳定性: 分子结构影响



Vacuum chamber  
40 C  
24 days

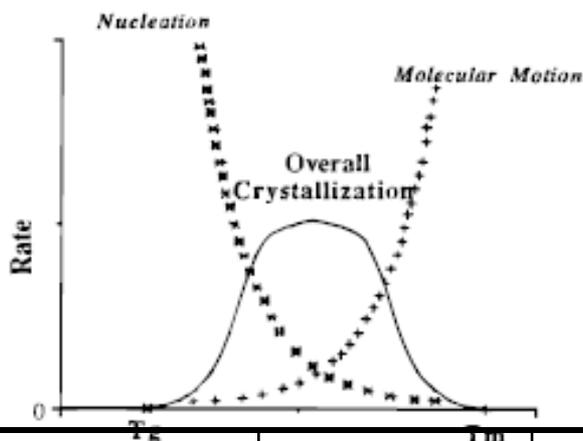


Vacuum chamber  
40 C  
4 days



# 结构弛豫是无定形稳定性的唯一参数？

Structure relaxation time is inversely proportional to the molecular mobility -  
A key determinant of the physical and/or chemical stability of amorphous phase



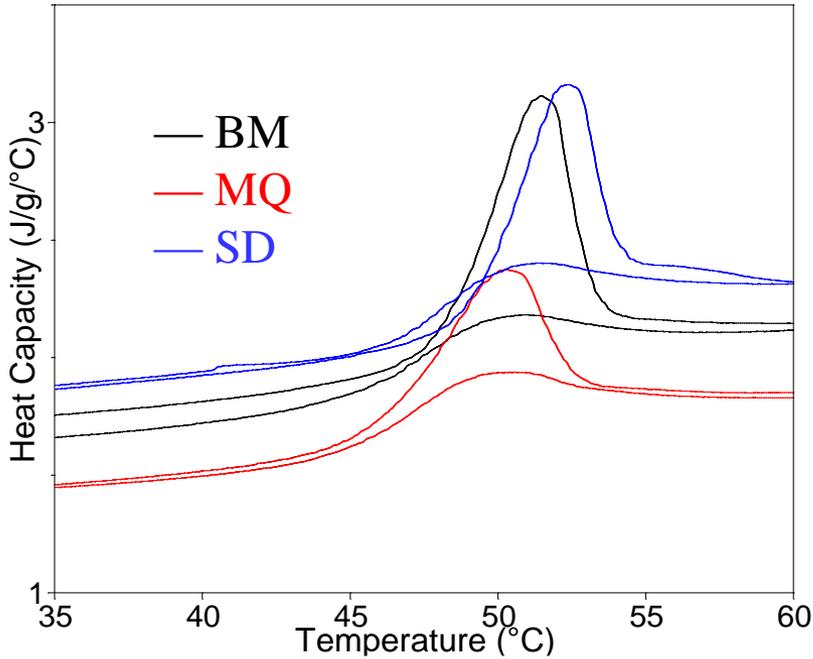
- Molecular mobility partially involved in the rate-limiting step for crystallization
- Extent of correlation between the rate of crystallization and molecular mobility could be different above and below  $T_g$

	$T_g$ (C)	$T_0$ (C)	m	D	$\tau_{40C}$ (s)	$\tau_{40C}$ (day)
<b>Compound#1</b>	44.5	-5	98	7.21	$5 \times 10^4$	0.5
<b>Compound#3</b>	78.5	24	104	6.69	$9 \times 10^{39}$	$10^{35}$

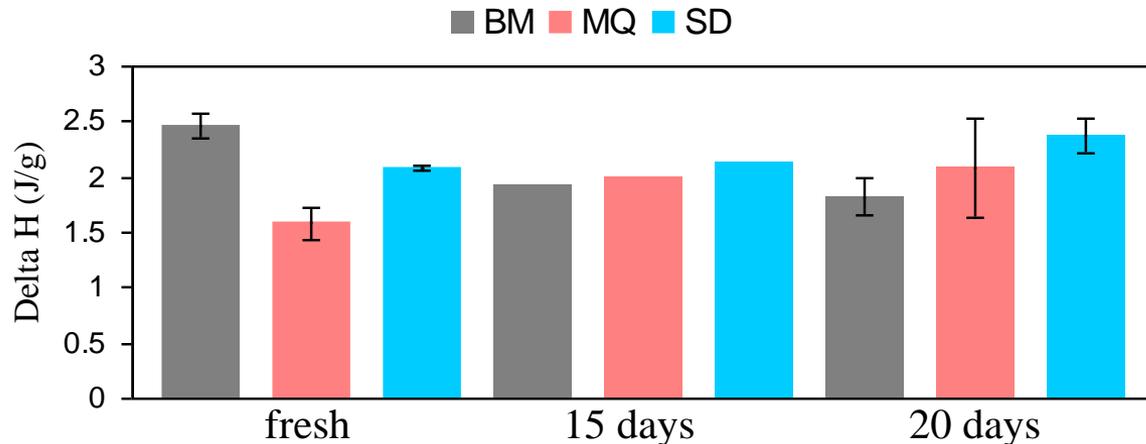
- Compound #1 with lower  $T_g$ , greater molecular mobility and faster dynamic shows much slower crystallization rate
- There are other important factors affect the crystallization process as well



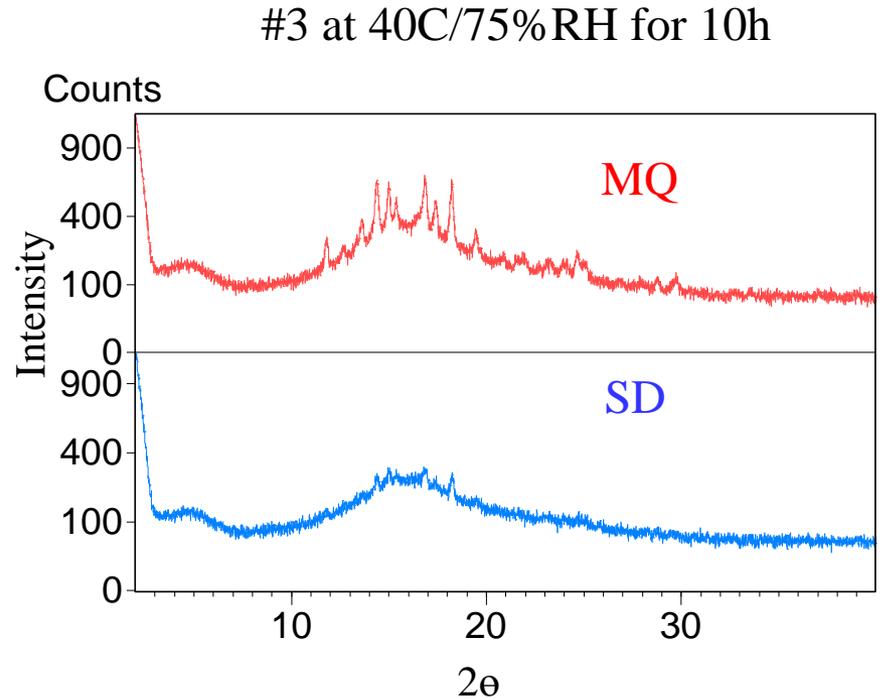
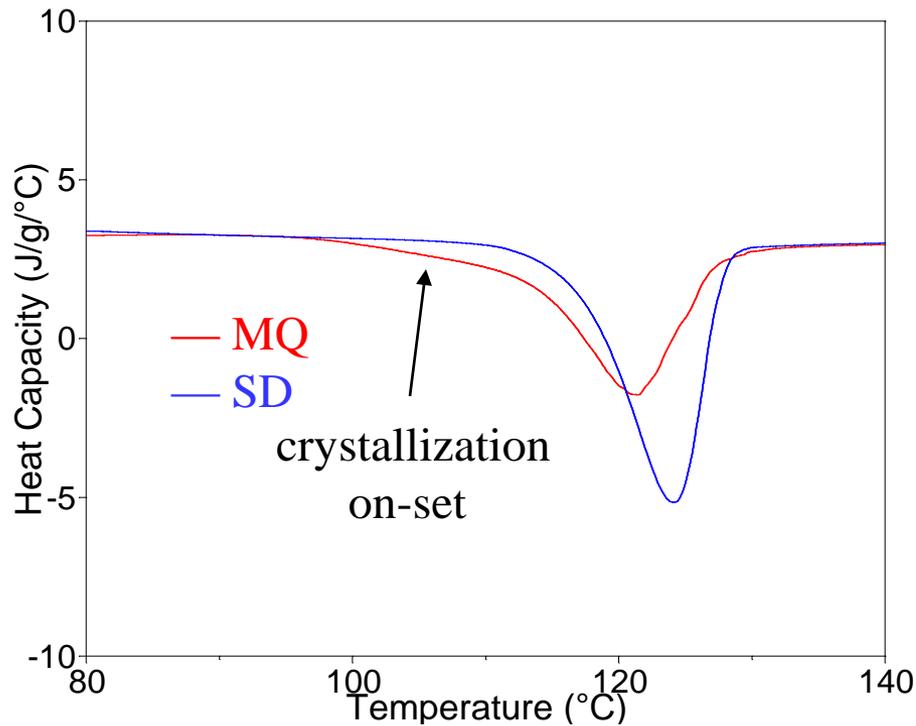
# 无定形制备方法对稳定性影响



- Different preparation methods result in varied excess enthalpy value (corresponding to different energy level in energy landscape)
- Enthalpy relaxes as a function of time



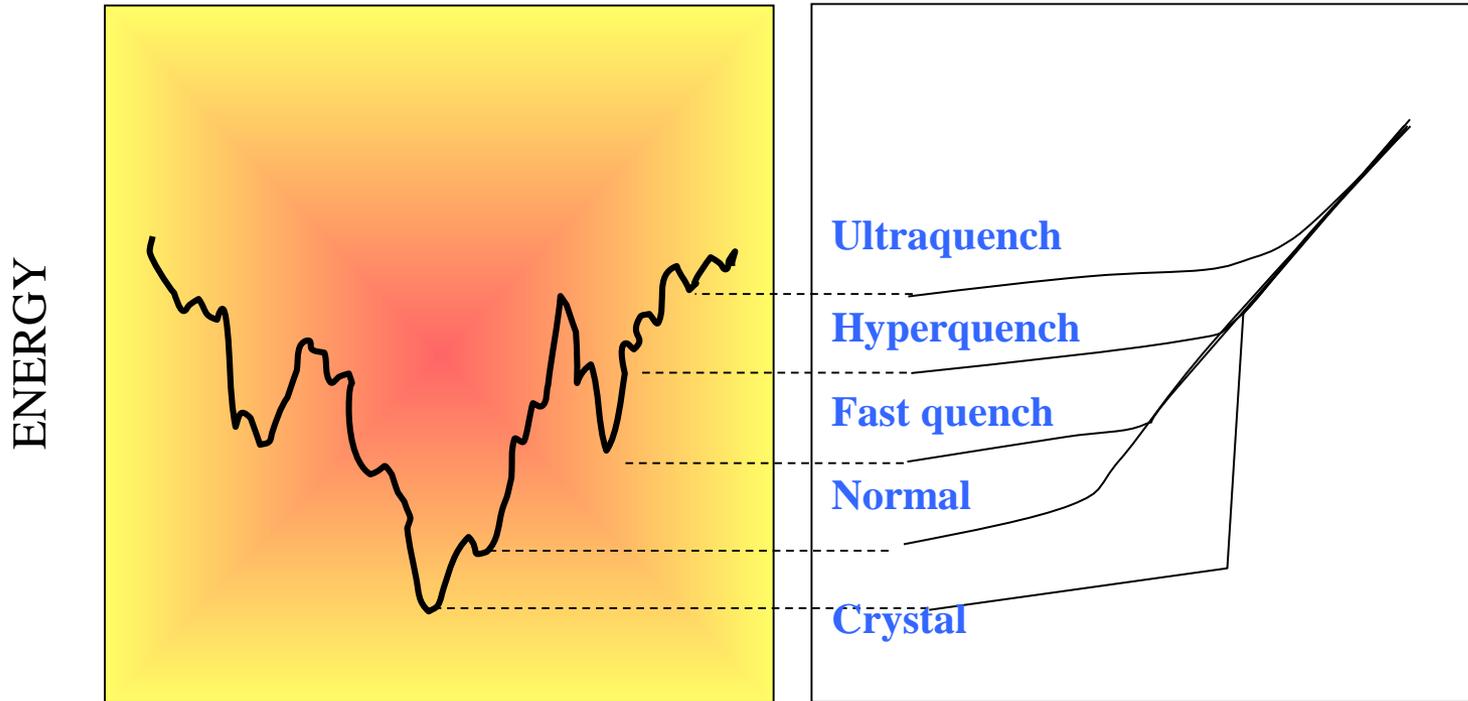
# 无定形制备方法对稳定性影响



- Spray dried (SD) #3 is more stable than melt quenched (MQ) #3.
- It is consistent with the result that SD freeze the system at a lower energy state (larger excess enthalpy).



# 用能量谱图对无定形稳定性的解释

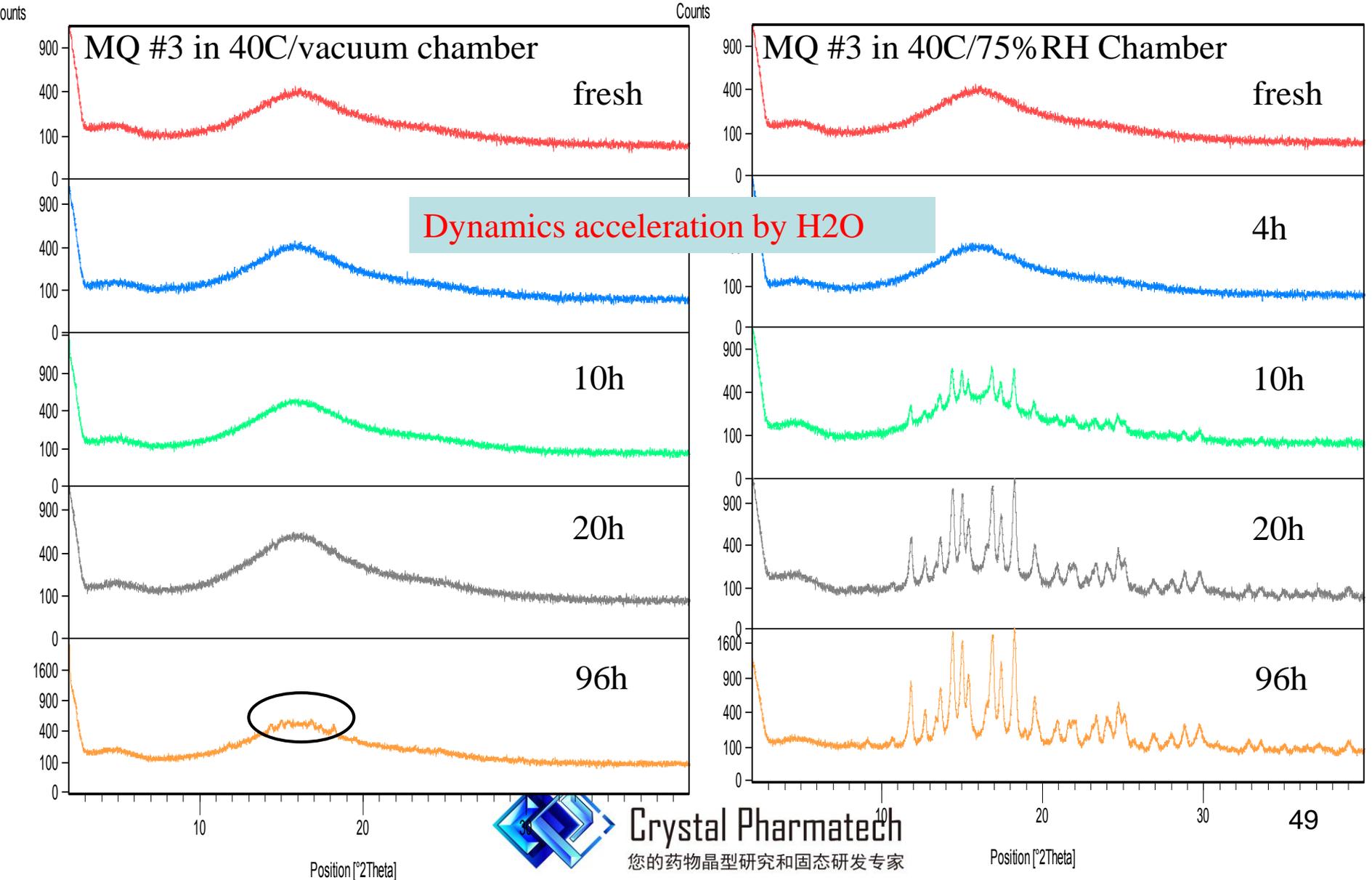


$Z^*$   $T$   
Relation between energy of glasses and quench rate

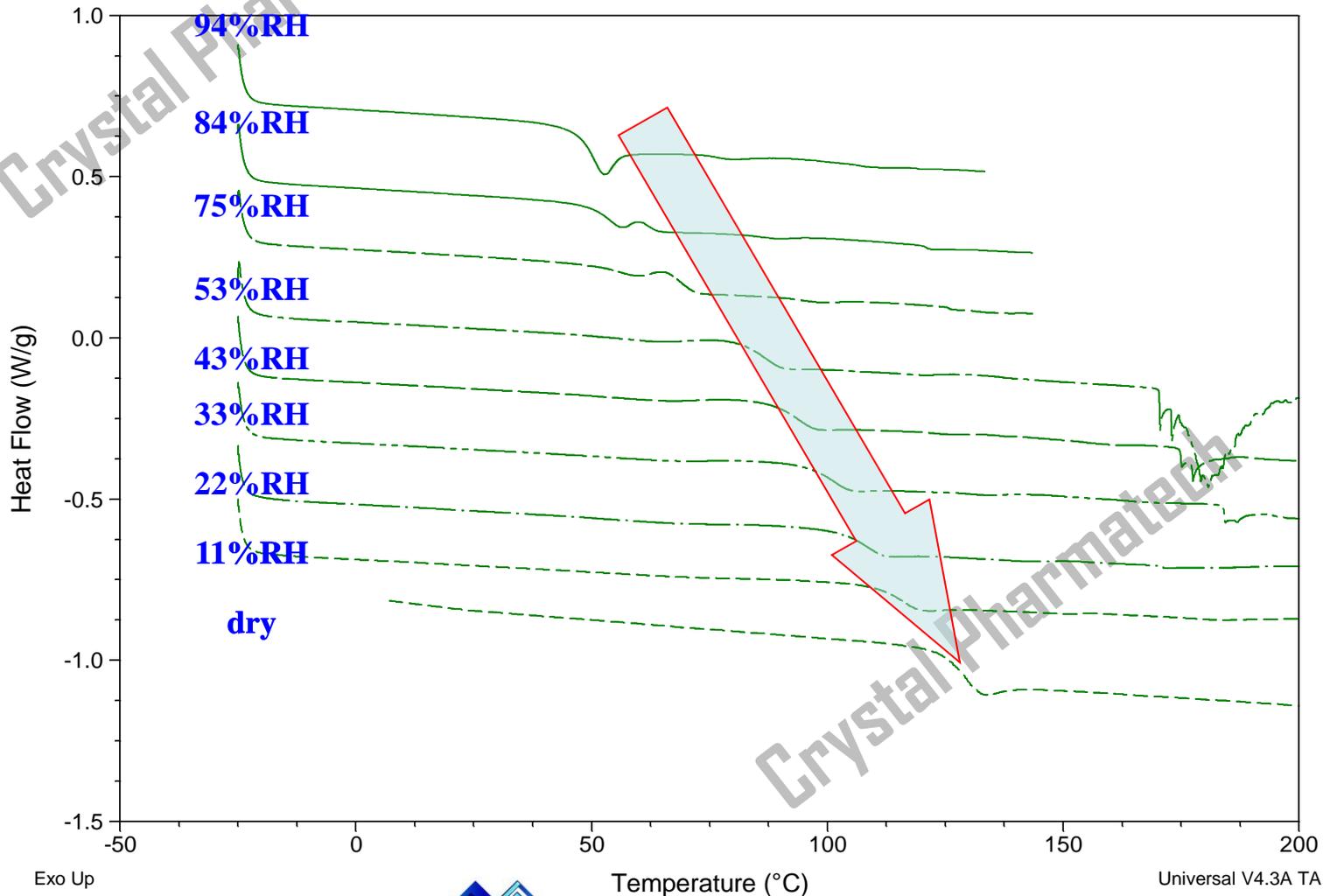
$$m \propto \frac{\text{Density of energy minima}}{\text{Energy barrier heights}}$$



# 水分对无定形稳定性影响



# 水分对玻璃化温度的影响



Exo Up

Temperature (°C)

Universal V4.3A TA Instruments

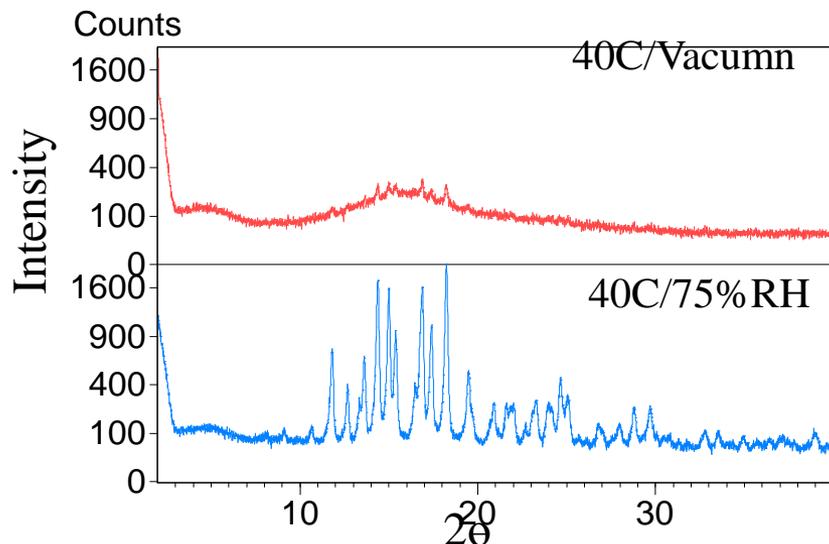


Crystal Pharmatech

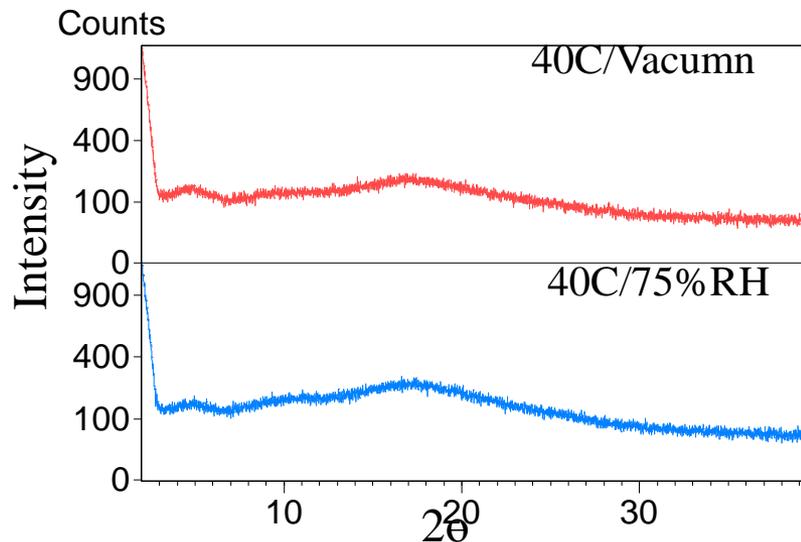
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# 高分子添加剂对无定形稳定性影响

Compound#3 (SD) 4 days



#3 + HPMCAS HF (SD, 1:2) 10 days



## Polymer stabilized API

- Slower molecular mobility?
- Changed nucleation process?



# 总结

- 无定形制剂研发越来越受重视，对溶解度较低 ( $D_0$  高) 或者晶型专利受保护的药物分子尤为重要
- 无定形的稳定性和许多物理性质密切相关，对无定形的深刻理解是开发稳定无定形制剂的关键
- 多种固态分析手段是无定形研发中必不可少的工具



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谢谢大家！

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